Studies on the Photochemical Rearrangements of Enantiomerically Pure, Polysubstituted, and Variously Annulated Bicyclo[2.2.2]octenones

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Supporting Information



ABSTRACT: A series of enantiomerically pure bicyclo [2.2.2] octenones, including the lactone-annulated system 26, has been prepared by engaging derivatives of an enzymatically derived and homochiral *cis*-1,2-dihydrocatechol in inter- or intra-molecular Diels–Alder reactions. Systems such as 26 readily participate in photochemically promoted oxa-di- π -methane rearrangement or 1,3-acyl migration processes to give products such as diquinane 34 or mixtures of cyclobutanone 36 and cyclopropane 38, respectively.

INTRODUCTION

Bicyclo[2.2.2] octenones including the parent system 1 (Scheme 1) are excellent substrates for certain photochemically promoted rearrangement reactions.¹ Specifically, on irradiation in the presence of photosensitizers such as acetophenone they participate in oxa-di- π -methane rearrangements and, so affording, via a triplet pathway, cyclopropannulated diquinanes such as 2. In contrast, on direct irradiation they engage, now via a singlet pathway, in a 1,3-acyl migration reaction (Givens rearrangement) to give bicyclo[4.2.0]oct-4-en-7-ones such as 3. Upon sustained irradiation, photoproduct 3 and many of its

Scheme 1. Oxa-di- π -methane, 1,3-Acyl Migration, and Photodecarbonylation Reactions of the Parent Bicyclo[2.2.2]octenone (1)



derivatives can undergo decarbonylation to give the corresponding $\Delta^2\text{-norcarene},$ e.g. 4.

The capacity to effectively exploit these valuable transformations is contingent on having ready access to bicyclo[2.2.2] octenones, especially enantiomerically pure ones. While various methods are available for the synthesis of such systems,² a particularly useful approach involves the facially selective intermolecular cycloaddition of dienophiles to enantiomerically pure *cis*-1,2-dihydrocatechols such as **5** (Figure 1) that are readily obtained through the whole-cell



Figure 1. Homochiral *cis*-1,2-dihydrocatechol 5 and the derived bicyclo[2.2.2]octenone-based photosubstrates 6 and 7.

biotransformation of the corresponding arene (toluene in the case of compound 5).³ Indeed, we have exploited such adducts in the photochemically mediated total synthesis of a range of sesquiterpenoid natural products including various triquinanes,⁴ protoilludanes,⁵ and the structure assigned to a sterpurene.⁶ In seeking to extend such studies for the purposes of preparing new molecular scaffolds, including those of use in drug discovery, we became interested in establishing the degree to which the bicyclo[2.2.2]octenone framework could be substituted and/or annulated and continue to engage in the above-

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mentioned photochemical processes. Herein, we detail research that serves to emphasize the remarkable extent to which these processes can be applied in a reliable fashion.

In connection with our recently reported studies⁷ on the synthesis of the sorbicillinoid-derived isolate rezishanone C, we prepared,⁷ using intermolecular Diels–Alder cycloaddition processes, the oxygenated bicyclo[2.2.2]octenones **6** and 7 in gram quantities from *cis*-1,2-dihydrocatechol **5**. Accordingly, compounds **6** and 7 became the substrates used in the opening stages of the present work.

RESULTS AND DISCUSSION

Independent irradiation (using a medium-pressure mercury lamp) of an acetone solution of each of compounds 6 and 7 held in a round-bottomed flask made from borosilicate glass and containing ca. 1.4 molar equiv of acetophenone (Scheme 2) afforded the anticipated oxa-di- π -methane rearrangement





products 8 (87% or 98% brsm) and 9 (90% or 99% brsm), respectively. All of the spectral data acquired on compounds 8 and 9 were entirely consistent with the assigned structures, but final confirmation of these followed from single-crystal X-ray analyses (see the Experimental Section and Supporting Information for details).

When dichloromethane solutions of substrates 6 and 7 were subjected to direct irradiation for brief periods (less than 1 h) with the same type of lamp then the anticipated Givens rearrangement products 10 (70% or 84% brsm) and 11 (67%), respectively, were obtained. Upon sustained irradiation (5 h) of the same substrates, significant quantities of the corresponding decarbonylated systems 12 (32%) and 13 (33%) were obtained along with reduced amounts of the corresponding and chromatographically separable cyclobutenones. Once again, all of the spectroscopic data acquired on compounds 10-13 were in accord with the assigned structures, but those of the first and the last of these were confirmed by single-crystal X-ray analysis.

Establishing the impact of various modes of ring-fusion on the capacity of bicyclo [2.2.2] octenones to engage in oxa-di- π -

methane and 1,3-acyl migration reactions was another topic of interest. Therefore, building upon our earlier studies⁸ on the assembly of the relevant frameworks through the intramolecular Diels-Alder (IMDA) reactions of crotonate esters derived from cis-1,2-dihydrocatechols, compound 5 was converted (Scheme 3) into the diesters 14 (81%) and 15 (74%) under previously established⁸ conditions. When each of compounds 14 and 15 was heated in refluxing toluene they engaged in the two possible modes of cycloaddition and thus giving rise to the corresponding pair of adducts, viz. lactones 16 (40% from 14), 17 (41% from 15), 18 (26% from 14), and 19 (12% from 15), that could be separated from one another by conventional chromatographic means. The remaining ester residues within each of adducts 18 and 19 were cleaved by standard methods and the resulting alcohols 20 (74%) and 21 (76%) oxidized to the corresponding ketones 22 (86%) and 23 (82%), respectively, using the Dess-Martin periodinane.⁹ Similarly, esters 16 and 17 were cleaved using potassium carbonate in methanol and the product alcohols 24 (79%) and 25 (94%), respectively, oxidized to the corresponding ketones, namely, compounds 26 (94%) and 27 (78%). Single-crystal X-ray analyses were conducted on compounds 23 and 27, details of which are presented in the Experimental Section and SI.

The oxa-di- π -methane rearrangement of the lactone-annulated bicyclo[2.2.2]octenones 22 and 23 took place readily when acetone solutions of each of these was irradiated in the presence of acetophenone (Scheme 4). By such means the cyclopropane-annulated oxatriquinanes 28 (40%) and 29 (43%) were obtained. In contrast, direct irradiation of substrates 22 and 23 led to the corresponding mixtures of cyclobutanones 30 (up to 75%) and 31 (up to 73%), respectively, as well as their decarbonylated counterparts 32 (up to 37%) and 33 (up to 52%), respectively. The structures of compounds 30 and 32 were confirmed by single-crystal Xray analyses.

Equivalent studies on the photochemical behaviors of the isomeric lactone annulated bicyclo[2.2.2]octenones 26 and 27 (Scheme 5) led to analogous outcomes. Specifically, the anticipated oxatriquinanes 34 (45%) and 35 (49%) were obtained on photosensitized irradiation of these substrates while direct irradiation afforded mixtures of cyclobutanones 36 (up to 70%) and 37 (up to 71%) as well as their decarbonylated congeners 38 (up to 35%) and 39 (up to 40%), respectively. The structures of compounds 34, 35, 36, and 38 were each confirmed by single-crystal X-ray analyses.

It is worth noting that both the substrates and the photoproducts shown in Scheme 5 are pseudoenantiomers of their counterparts shown in Scheme 4. Thus, for example, if each of compounds **30** (Scheme 4) and **36** (Scheme 5) lacked the two methyl groups then they would be enantiomers. Accordingly, the protocols just described provide a means by which the single enantiomeric form of starting material **5** can be converted into either enantiomeric form of a range of relatively complex molecular frameworks.

In seeking to establish the effects of alternate modes of ring fusion and other substituents on the capacity of bicyclo[2.2.2] octenones to engage in photochemically promoted rearrangements, the behaviors of various polyhydro-4,7-ethanoindenone derivatives were explored. The routes shown in Scheme 6 were used to prepare these cyclopentannulated systems. Thus, the previously described^{4d} tricyclic system 40, prepared by a reaction sequence involving an initial Diels–Alder cyclo-addition reaction between 2-cyclopenten-1-one and the



Scheme 4. Photochemical Behaviors of Bicyclo[2.2.2]octenones 22 and 23 under Either Direct Irradiation or Sensitized Conditions



acetonide derived from *cis*-diol **5**, was reduced with LiAlH₄, and the previously reported^{4d,5b} alcohol **41** (39%) obtained in pure form after chromatographic separation from its coproduced epimer.^{4d,5b} Compound **41**, the structure of which was confirmed by single-crystal X-ray analysis, was acetylated under standard conditions to give ester **42** (97%), and the acetonide moiety associated with this last compound was hydrolyzed by treating a methanol/water solution of it with acidified AG-50W-X8 resin. The product diol **43** (97%) was selectively oxidized using the sterically demanding oxammoScheme 5. Photochemical Behaviors of Bicyclo[2.2.2]octenones 26 and 27 under Either Direct Irradiation or Sensitized Conditions



nium salt derived from *p*-TsOH·H₂O-promoted disproportionation of 4-acetamido-TEMPO,¹⁰ and thus producing acyloin 44 (97%) that was also subjected to single-crystal X-ray analysis. Compound 44 was converted into derivatives 45 (91%), 46 (77%), 47 (82%), and 48 (93%) by standard methods, while samarium iodide mediated deoxygenation of benzoate 48 also allowed for the preparation of bicyclo[2.2.2]octenone 49 (85%). Once again, the structures of compounds 45, 47, and 48 were confirmed by single-crystal X-ray analyses.

Scheme 6. Synthesis of Cyclopentannulated Bicyclo[2.2.2]octenones 45–49 and Their Photochemical Behaviors under Direct Irradiation Conditions



On direct irradiation of dichloromethane solutions of each of compounds 45-49 they engaged in 1,3-acyl migration reactions to give the cyclobutanones 50 (43% or 99% brsm), 51 (27% or 98% brsm), 52 (18% or 99% brsm), 53 (17% or quant brsm), and 54 (23% or 92% brsm), respectively. Irradiation of compounds 50 and 47 led to cyclopropanes 55 (98%) and 56 (80%), respectively. The structures of all of these

photoproducts were determined through comprehensive spectroscopic analyses. In particular, the infrared spectrum of each of cyclobutanones **50–54** exhibited a characteristic carbonyl absorption band in the range 1778–1792 cm⁻¹ as well as a carbonyl carbon resonance above δ 200 ppm in the corresponding ¹³C NMR spectrum. The structure of compound **53** was confirmed by single-crystal X-ray analysis.

Given our extensive earlier studies⁴ on the oxa-di- π -methane rearrangements of substrates very closely related to compounds **45–49**, we have not subjected these to photosensitized irradiation but would fully expect them to behave in a similar manner and thus leading to the corresponding cyclopropannulated triquinanes.

In contrast to outcomes noted above (Scheme 6), when the mesylate 57 (Scheme 7), which was readily prepared in 56%





yield from acyloin 44, was subjected (as a solution in dichloromethane) to direct irradiation it engaged in an oxadi- π -methane rearrangement reaction rather than a 1,3-acyl migration reaction and so affording the cyclopentannulated and stereochemically pure triquinane 58 in 48% yield (or 98% yield brsm). The configuration at the carbon bearing the mesyloxy group is tentatively assigned as illustrated although the alternate one cannot be discounted as a result of the intervention of a photoepimerization process.^{4b-d} The precise origins of this seemingly anomalous behavior of sulfonate ester 57 remain unclear at the present time and serve to highlight the very limited understanding of the photochemical properties of such systems.¹¹ These matters are the subject of ongoing studies in our laboratories.

CONCLUSIONS

The studies reported here have established that a wide range of extensively substituted and variously annulated bicyclo[2.2.2]-octenones are capable of engaging in either oxa-di- π -methane or 1,3-acyl migration reactions depending upon the mode of photoactivation (sensitized vs direct irradiation). The 1,3-acyl migration reactions leading to the isomeric cyclobutanones can be followed by rapid photodecarbonylation processes to

coproduce the corresponding (but normally readily separable) cyclopropanes. A noteworthy feature of the photochemical behaviors of the annulated bicyclo[2.2.2]octenones studied here is that the mode of ring fusion (annulation) can have a significant impact on the rates of both types of processes. Thus, for example, the lactone-annulated systems 22, 23, 26, and 27 appear to engage in more rapid 1,3-acyl migration reactions than their cyclopentannulated counterparts 45-49. The same could be said of the corresponding oxa-di- π -methane process, especially when the outcomes of our earlier studies⁴ are taken into account. That is to say, the lactone annulated compounds just mentioned also appear to participate in more rapid tripletsensitized rearrangement reactions than their cyclopentannulated counterparts. The origins of these variations are the subject of ongoing studies, the outcomes of which will be reported in due course.

EXPERIMENTAL SECTION

General Experimental Procedures. Unless otherwise specified, proton (¹H) and carbon (¹³C) NMR spectra were recorded at 18 °C in base-filtered CDCl₃ on a spectrometer operating at 400 MHz for proton and 100 MHz for carbon nuclei. ¹H NMR data are recorded as follows: chemical shift (δ) [multiplicity, coupling constant(s) J (Hz), relative integral] where multiplicity is defined as s = singlet; d = doublet; t = triplet; q = quartet; m = multiplet or combinations of the above. In relevant cases, the signal due to residual CHCl₃ appearing at $\delta_{\rm H}$ 7.26 and the central resonance of the CDCl₃ "triplet" appearing at $\delta_{\rm C}$ 77.0 were used to reference ¹H and ¹³C NMR spectra, respectively. Samples were analyzed by infrared spectroscopy (ν_{max}) as thin films on KBr plates. Optical rotations were recorded using the sodium D-line (589 nm) in a cell with a path length of 1 dm, at the concentrations indicated and in the specified solvent at 22 °C. Specific rotations were then calculated in the usual manner. Low- and high-resolution electron impact (EI) mass spectra were recorded on a double focusing, triple sector machine. Low- and high-resolution ESI mass spectra were recorded on a triple-quadrupole mass spectrometer operating in positive ion mode. Melting points are uncorrected. Analytical thin layer chromatography (TLC) was performed on aluminum-backed 0.2 mm thick silica gel 60 F₂₅₄ plates. Eluted plates were visualized using a 254 nm UV lamp and/or by treatment with a suitable dip followed by heating. These dips included phosphomolybdic acid/ceric sulfate/ sulfuric acid (conc)/water (37.5 g: 7.5 g: 37.5 g: 720 mL), potassium permanganate/potassium carbonate/5% sodium hydroxide aqueous solution/water (3 g: 20 g: 5 mL: 300 mL), and p-anisaldehyde or vanillin/sulfuric acid (conc)/ethanol (15 g: 2.5 mL: 250 mL). Flash chromatographic separations were carried out following protocols defined by Still et al.¹² with silica gel 60 (40–63 μ m) as the stationary phase and using the AR- or HPLC-grade solvents indicated. The melting points of solids purified by such means were recorded directly (ie after they had crystallized from the concentrated chromatographic fractions). Starting materials, reagents, drying agents, and other inorganic salts were generally commercially available and were used as supplied. Tetrahydrofuran (THF), methanol, and dichloromethane were dried using a solvent purification system that is based upon a technology originally described by Grubbs et al.¹³ Where necessary, reactions were performed under a nitrogen atmosphere.

Specific Chemical Transformations. ($2a'R, 3\bar{S}, 4S, 4aR$)-3, 4-Dihydroxy-2b-methylhexahydrocyclopropa[cd]pentalen-2(1H)-one (8). A deoxygenated solution of compound 6⁷ (94 mg, 0.56 mmol) and acetophenone (90 μ L, 0.77 mmol) in acetone (100 mL) maintained under nitrogen was subjected to irradiation with a Hanovia 450 W medium-pressure mercury-vapor lamp for 2.5 h and then cooled and concentrated under reduced pressure to give a yellow oil. This material was subjected to flash column chromatography (silica, 1:1 \rightarrow 1:0 v/v ethyl acetate/40–60 petroleum ether gradient elution) to afford two fractions, A and B. Concentration of fraction A ($R_f = 0.3$ in ethyl acetate) afforded the starting compound 6 (10 mg, 11% recovery) as a colorless, crystalline solid that was identical, in all respects, with an authentic sample.

Concentration of fraction B ($R_f = 0.1$ in ethyl acetate) afforded compound **8** (82 mg, 87% or 98% brsm) as a white, crystalline solid: mp = 169–171 °C; [α]_D = +71.4 (*c* 1.0, methanol); ¹H NMR (400 MHz, CD₃OD) δ 4.15 (d, *J* = 2.6 Hz, 1H), 3.74 (s, 1H), 2.81 (t, *J* = 5.0 Hz, 1H), 2.72 (m, 1H), 2.42 (m, 1H), 2.08 (m, 1H), 1.82 (d, *J* = 5.0 Hz, 1H), 1.37 (s, 3H) (signals due to O–H group protons not observed); ¹³C NMR (100 MHz, CD₃OD) δ 217.3, 88.2, 86.3, 47.8, 47.0, 46.3(0), 46.2(8), 44.2, 21.5; IR ν_{max} 3356, 2962, 2904, 1706, 1693, 1290, 1040, 898 cm⁻¹; MS (ESI, + ve) *m*/*z* 191 [(M + Na)⁺, 100]; HRMS *m*/*z* (M + Na)⁺ calcd for C₉H₁₂O₃Na 191.0684, found 191.0685.

(2a' R,3S,4S,4aR)-4-((tert-Butyldimethylsilyl)oxy)-3-hydroxy-2bmethylhexahydrocyclopropa[cd]pentalen-2(1H)-one (9). A deoxygenated solution of compound 7^7 (158 mg, 0.56 mmol) and acetophenone (90 μ L, 0.77 mmol) in acetone (100 mL) maintained under nitrogen was subjected to irradiation with a Hanovia 450 W medium-pressure mercury-vapor lamp for 2.5 h. The cooled reaction mixture was concentrated under reduced pressure to give a clear, yellow oil, and this was subjected to flash column chromatography (silica, 1:9 \rightarrow 3:7 v/v ethyl acetate/40–60 petroleum ether gradient elution) to afford two fractions, A and B.

Concentration of fraction A ($R_f = 0.4$ in 3:7 v/v ethyl acetate/40–60 petroleum ether) afforded compound 7 (13 mg, 8% recovery) as a colorless crystalline solid that was identical, in all respects, with an authentic sample.

Concentration of fraction B ($R_f = 0.2$ in 3:7 v/v ethyl acetate/40– 60 petroleum ether) afforded compound 9 (143 mg, 90% or 99% brsm) as a white, crystalline solid: mp = 99–100 °C; [α]_D = +41.8 (c1.0, CHCl₃); ¹H NMR (800 MHz, CDCl₃) δ 4.20 (d, J = 2.5 Hz, 1H), 3.81 (s, 1H), 2.76 (t, J = 5.1 Hz, 1H), 2.66 (m, 1H), 2.38 (m, 1H), 2.13 (d, J = 17.9 Hz, 1H), 1.86 (d, J = 5.0 Hz, 1H), 1.79 (broad s, 1H), 1.35 (s, 3H), 0.90 (s, 9H), 0.08 (s, 6H); ¹³C NMR (200 MHz, CDCl₃) δ 213.3, 86.8, 85.8, 45.3, 45.0(4), 44.9(8), 44.4, 42.2, 24.9, 20.4, 17.2, – 5.6; IR ν_{max} 3351, 2957, 2929, 2886, 2857, 1719, 1254, 1092, 867, 836, 771 cm⁻¹; MS (ESI, + ve) m/z 305 [(M + Na)⁺, 100], 283 [(M + H)⁺, 30]; HRMS m/z (M + Na)⁺ calcd for C₁₅H₂₆O₃SiNa 305.1549, found 305.1548.

(1R,2S,3S,6R)-2,3-Dihydroxy-4-methylbicyclo[4.2.0]oct-4-en-7one (10). A magnetically stirred and deoxygenated solution of compound 6 (100 mg, 0.59 mmol) in dichloromethane (100 mL) maintained under nitrogen was subjected to direct irradiation with a Hanovia 450 W medium-pressure mercury-vapor lamp for 0.83 h. The cooled reaction mixture was concentrated under reduced pressure to give a yellow solid that was subjected to flash column chromatography (silica, 2:3 \rightarrow 1:0 v/v ethyl acetate/40–60 petroleum ether gradient elution) to afford two fractions, A and B.

Concentration of fraction A ($R_f = 0.4$ in ethyl acetate) afforded compound **10** (70 mg, 70% or 84% brsm) as a white, crystalline solid: mp = 83–85 °C; [α]_D = -234.4 (*c* 1.0, MeOH); ¹H NMR (400 MHz, CD₃OD) δ 5.40 (m, 1H), 3.96 (d, *J* = 6.9 Hz, 1H), 3.88 (m, 1H), 3.47 (t, *J* = 7.5 Hz, 1H), 3.24 (m, 1H), 2.98 (m, 1H), 2.57 (m, 1H), 1.81 (s, 3H) (signals due to O–H group protons not observed); ¹³C NMR (100 MHz, CD₃OD) δ 209.0, 140.2, 116.7, 76.2, 73.6, 61.0, 50.1, 30.2, 20.0; IR ν_{max} 3391, 2882, 1769, 1066, 1047, 1014, 884 cm⁻¹; MS (ESI, + ve) *m*/*z* 223 [(M + MeOH + Na)⁺, 40], 191 [(M + Na)⁺, 100]; HRMS *m*/*z* (M + Na)⁺ calcd for C₉H₁₂O₃Na 191.0684, found 191.0684.

Concentration of fraction B ($R_f = 0.3$ in ethyl acetate) afforded compound 6 (14 mg, 14% recovery) as a colorless, crystalline solid that was identical, in all respects, with an authentic sample.

(1R,2S,3S,6R)-2-((tert-Butyldimethylsilyl)oxy)-3-hydroxy-4methylbicyclo[4.2.0]oct-4-en-7-one (11). A magnetically stirred and deoxygenated solution of compound 7 (100 mg, 0.35 mmol) in dichloromethane (100 mL) maintained under nitrogen was subjected to direct irradiation with a Hanovia 450 W medium-pressure mercuryvapor lamp for 0.67 h. The cooled reaction mixture was concentrated under reduced pressure to give a pale-yellow oil. This was subjected to flash column chromatography (silica, 1:9 v/v ethyl acetate/40–60 petroleum ether elution) to give, after concentration of the appropriate fractions ($R_f = 0.6$ in 3:7 v/v ethyl acetate/40–60 petroleum ether), compound **11** (67 mg, 67%) as a white, crystalline solid: mp = 67–69 °C; [α]_D = -139.6 (*c* 1.0, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ 5.38 (m, 1H), 4.04 (d, *J* = 7.0 Hz, 1H), 3.86 (m, 1H), 3.57 (t, *J* = 7.7 Hz, 1H), 3.24 (m, 1H), 2.94 (m, 1H), 2.57 (m, 1H), 2.18 (broad s, 1H), 1.82 (s, 3H), 0.90 (s, 9H), 0.13 (s, 3H), 0.10 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 207.0, 138.4, 115.1, 77.9, 73.4, 60.6, 50.2, 29.9, 25.9, 19.1, 18.3, – 3.8, – 4.2; IR ν_{max} 3499, 2954, 2930, 2857, 1779, 1126, 1087, 836 cm⁻¹; MS (ESI, + ve) *m*/*z* 337 [(M + MeOH + Na)⁺, 30], 305 [(M + Na)⁺, 100]; HRMS *m*/*z* (M + Na)⁺ calcd for C₁₅H₂₆O₃SiNa 305.1549, found 305.1554.

(1R,2S,3S,6R)-4-Methylbicyclo[4.1.0]hept-4-ene-2,3-diol (12). A magnetically stirred and deoxygenated solution of compound 6 (100 mg, 0.59 mmol) in dichloromethane (100 mL) maintained under a nitrogen atmosphere was subjected to direct irradiation with a Hanovia 450 W medium-pressure mercury-vapor lamp for 5 h. The cooled reaction mixture was concentrated under reduced pressure to give a brown oil that was subjected to flash column chromatography (silica, $1:4 \rightarrow 1:1 \text{ v/v}$ ethyl acetate/40–60 petroleum ether gradient elution) to afford two fractions, A and B.

Concentration of fraction A ($R_f = 0.6$ in ethyl acetate) afforded compound **12** (27 mg, 32%) as a pale-yellow oil: $[\alpha]_D = +93.2$ (*c* 0.8, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ 6.00 (m, 1H), 4.04–3.99 (complex m, 2H), 1.92 (broad s, 2H), 1.76 (s, 3H), 1.48–1.35 (complex m, 2H), 1.03 (m, 1H), 0.64 (m, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 130.3, 127.8, 73.3, 70.1, 21.4, 19.6, 18.7, 11.0; IR ν_{max} 3367, 2920, 1445, 1260, 1048, 1001, 724 cm⁻¹; MS (EI, 70 eV) *m/z* 140 (M^{•+}, 30), 122 [(M – H₂O)^{•+}, 60], 93 (100), 79 (90), 77 (70), 70 (60), 55 (60); HRMS *m/z* M^{•+} calcd for C₈H₁₂O₂ 140.0837, found 140.0842.

Concentration of fraction B ($R_f = 0.4$ in ethyl acetate) afforded compound **10** (48 mg, 48%) as a white, crystalline solid that was identical, in all respects, with an authentic sample.

(1R,2S,3S,6R)-2-((tert-Butyldimethylsilyl)oxy)-4-methylbicyclo-[4.1.0]hept-4-en-3-ol (13). A magnetically stirred and deoxygenated solution of compound 7 (100 mg, 0.35 mmol) in dichloromethane (100 mL) maintained under a nitrogen atmosphere was subjected to irradiation with a Hanovia 450 W medium-pressure mercury-vapor lamp for 5 h. The cooled reaction mixture was concentrated under reduced pressure to give a brown oil that was subjected to flash column chromatography (silica, 1:9 \rightarrow 1:4 v/v ethyl acetate/40–60 petroleum ether gradient elution) to afford two fractions, A and B.

Concentration of fraction A ($R_f = 0.7$ in 3:7 v/v ethyl acetate/40– 60 petroleum ether) afforded compound **13** (30 mg, 33%) as a colorless, crystalline solid: mp = 32–33 °C; $[\alpha]_D = +35.3$ (*c* 1.0, CHCl₃). ¹H NMR (400 MHz, CDCl₃) δ 5.71 (m, 1H), 4.08 (d, *J* = 5.9 Hz, 1H), 3.59 (m, 1H), 1.97 (broad s, 1H), 1.74 (s, 3H), 1.33–1.17 (complex m, 2H), 0.99–0.94 (complex m, 1H), 0.92 (s, 9H), 0.32 (m, 1H), 0.14 (s, 3H), 0.12 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 134.4, 124.7, 75.1, 74.0, 26.0, 19.6, 19.3, 18.7, 18.3, 12.0, – 4.1, – 4.6; IR ν_{max} 3458, 2929, 2857, 1254, 1075, 1003, 835, 775 cm⁻¹; MS (EI, 70 eV) *m*/*z* 239 [(M – CH₃•)⁺, 30], 211 (35), 197 [(M – C₄H₉•)⁺, 35], 105 (60), 75 (100), 73 (95); HRMS *m*/*z* (M – CH₃•)⁺ calcd for C₁₃H₂₃O₂Si 239.1467, found 239.1468.

Concentration of fraction B ($R_f = 0.6$ in 3:7 v/v ethyl acetate/40– 60 petroleum ether) afforded compound 11 (33 mg, 33%) as a colorless, crystalline solid that was identical, in all respects, with an authentic sample.

(15,2R)-3-Methylcyclohexa-3,5-diene-1,2-diyl (2E,2'E)-Bis(but-2enoate) (14). A magnetically stirred solution of (1S,2R)-3-methylcyclohexa-3,5-diene-1,2-diol (5) (2.40 g, 19.0 mmol) in dry THF (284 mL) was cooled to -78 °C and then treated, dropwise over 0.5 h, with *n*-BuLi (12.5 mL of a 1.6 M solution in hexane, 20.0 mmol). The resulting solution was stirred at -78 °C for 0.17 h and then treated with crotonoyl chloride (1.80 mL, 19.02 mmol) over 0.33 h. The resulting solution was stirred for 2 h at -78 °C and then treated, dropwise over 0.5 h, with *n*-BuLi (12.5 mL of a 1.6 M solution in hexane, 20.0 mmol). After 0.17 h, another portion of crotonoyl chloride (1.80 mL, 19.02 mmol) was added over 0.33 h, and then stirring of the recation mixture was continued at -78 °C for 1 h. The ensuing mixture was warmed to 22 °C and then quenched with NaHCO₃ (96 mL of a saturated aqueous solution) before being diluted with diethyl ether (390 mL). The separated organic phase was washed with NH₄Cl (1 × 96 mL of a saturated aqueous solution) and then dried (MgSO₄), filtered, and concentrated under reduced pressure. The residue thus obtained was subjected to flash column chromatography to give, after concentration of the appropriate fractions (R_f = 0.5 in 1:4 v/v ethyl acetate/40–60 petroleum ether), compound **14** (4.04 g, 81%) as a pale-yellow oil. The spectral data recorded on this material matched those reported previously.⁸

(1S,2R)-3-Methylcyclohexa-3,5-diene-1,2-divl (2E,2'E)-Bis(4methylpent-2-enoate) (15). A magnetically stirred solution of (1S,2R)-3-methylcyclohexa-3,5-diene-1,2-diol (5) (2.40 g, 19.02 mmol) in dry THF (284 mL) was cooled to -78 °C and then treated, dropwise over 0.5 h, with *n*-BuLi (12.5 mL of a 1.6 M solution in hexane, 20.0 mmol). The resulting solution was stirred at -78 °C for 0.17 h and then treated, dropwise over 0.33 h, with (E)-4methylpent-2-enoyl chloride¹⁴ (2.50 mL, 19.02 mmol). The ensuing mixture was stirred at -78 °C for 2 h and again treated, dropwise over 0.5 h, with n-BuLi (12.5 mL of a 1.6 M solution in hexane, 20.0 mmol). After 0.17 h, further (E)-4-methylpent-2-enoyl chloride (1.80 mL, 19.02 mmol) was added, again dropwise over 0.33 h, and stirring was continued at $-78\ ^\circ C$ for 1 h. The reaction mixture was then allowed to warm to 22 °C before being quenched with NaHCO3 (96 mL of a saturated aqueous solution) and diluted with diethyl ether (390 mL). The separated organic phase was washed with NH_4Cl (1 × 96 mL of a saturated aqueous solution), dried (MgSO₄), filtered, and concentrated under reduced pressure. The residue thus obtained was subjected to flash column chromatography to give, after concentration of the appropriate fractions ($R_f = 0.6$ in 1:4 v/v ethyl acetate/40–60 petroleum ether), compound 15 (4.47 g, 74%) as a pale-yellow oil: $[\alpha]_{\rm D}$ = +71.6 (c 1.0, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ 6.98– 6.88 (complex m, 2H), 6.06 (m, 1H), 5.88 (broadened d, J = 5.2 Hz, 1H), 5.81-5.72 (complex m, 3H), 5.62 (s, 2H), 2.43 (m, 2H), 1.84 (s, 3H), 1.05–1.02 (complex m, 12H); ¹³C NMR (100 MHz, CDCl₃) δ 166.5, 166.3, 156.5, 156.2, 134.7, 126.6, 123.0, 121.9, 118.3, 118.1, 69.5, 68.0, 31.1, 31.0, 21.3, 21.2, 19.8 (two signals obscured or overlapping); IR v_{max} 2963, 1717, 1651, 1297, 1260, 1157, 982, 859 cm^{-1} ; MS (ESI, + ve) m/z 341 [(M + Na)⁺, 100]; HRMS m/z: (M + Na)⁺ calcd for C₁₉H₂₆O₄Na 341.1729, found 341.1728.

3a,8-Dimethyl-2-oxo-2,3,3a,6,7,7a-hexahydro-3,6-methanobenzofuran-7-yl (E)-But-2-enoate (16) and 6,8-Dimethyl-2-oxo-2,3,3a,6,7,7a-hexahydro-3,6-methanobenzofuran-7-yl (E)-But-2enoate (18). A magnetically stirred solution of compound 14 (4.04 g, 15.40 mmol) in toluene (97 mL) maintained under a nitrogen atmosphere was heated at reflux for 24 h then cooled to 22 °C and concentrated under reduced pressure. The resulting yellow oil was subjected to flash chromatography (silica, 1:19 v/v ethyl acetate/40– 60 petroleum ether elution) to give two fractions, A and B.

Concentration of fraction A $[R_f = 0.2(5)$ in 1:4 v/v ethyl acetate/ 40–60 petroleum ether] afforded compound **16** (1.62 g, 40%) as a white, crystalline solid, mp = 45–47 °C. The spectral data recorded on this material matched those reported previously.⁸

Concentration of fraction B ($R_f = 0.2$ in 1:4 v/v ethyl acetate/40–60 petroleum ether) afforded compound **18** (1.06 g, 26%) as a clear, colorless oil. The spectral data recorded on this material matched those reported previously.⁸

8-Isopropyl-3a-methyl-2-oxo-2,3,3a,6,7,7a-hexahydro-3,6-methanobenzofuran-7-yl (E)-4-Methylpent-2-enoate (17) and 8-Isopropyl-6-methyl-2-oxo-2,3,3a,6,7,7a-hexahydro-3,6-methanobenzofuran-7-yl (E)-4-Methylpent-2-enoate (19). A magnetically stirred solution of compound 15 (4.47 g, 14.04 mmol) in toluene (88 mL) maintained under a nitrogen atmosphere was heated at reflux for 72 h then cooled to 22 °C, and concentrated under reduced pressure. The resulting yellow oil was subjected to flash chromatography (silica, 1:19 v/v ethyl acetate/40–60 petroleum ether elution) to afford two fractions, A and B.

Concentration of fraction A ($R_f = 0.4$ in 1:4 v/v ethyl acetate/40–60 petroleum ether) afforded compound 17 (1.83 g, 41%) as a white, crystalline solid: mp = 69–71 °C; [α]_D = -115.4 (c 1.0, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ 6.97 (dd, J = 15.7 and 6.5 Hz, 1H), 6.38 (t, J = 7.6 Hz, 1H), 5.84 (m, 2H), 4.37 (dd, J = 6.7 and 2.2 Hz, 1H), 4.24 (d, J = 6.7 Hz, 1H), 2.98 (m, 1H), 2.47 (m, 1H), 1.93–1.88 (complex m, 2H), 1.40 (s, 3H), 1.27 (m, 1H), 1.08 (s, 3H), 1.06 (s, 3H), 0.97 (d, J = 6.5 Hz, 3H), 0.90 (d, J = 6.6 Hz, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 179.3, 166.5, 156.8, 133.1, 132.2, 117.9, 77.3, 70.6, 49.7, 46.9, 45.1, 38.8, 31.1, 30.5, 21.3, 21.2, 20.9, 20.8, 20.7; IR ν_{max} 2962, 1783, 1717, 1264, 1160, 1038, 1009, 902 cm⁻¹; MS (EI, 70 eV) m/z 318 (M^{•+}, 5), 156 (15), 133 (10), 97 (100), 69 (10), 41 (20); HRMS m/z M^{•+} calcd for C₁₉H₂₆O₄ 318.1831, found 318.1834.

Concentration of fraction B [$R_f = 0.3(5)$ in 1:4 v/v ethyl acetate/ 40–60 petroleum ether] afforded compound **19** (537 mg, 12%) as a clear, colorless oil: [α]_D = +176.0 (*c* 1.0, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ 6.97 (dd, *J* = 15.7 and 6.5 Hz, 1H), 6.09 (d, *J* = 8.2 Hz, 1H), 6.00 (dd, *J* = 8.2 and 6.6 Hz, 1H), 5.83 (dd, *J* = 15.7 and 1.5 Hz, 1H), 4.57 (m, 1H), 4.05 (d, *J* = 6.7 Hz, 1H), 3.45 (m, 1H), 2.47 (m, 1H), 2.30 (m, 1H), 2.16 (m, 1H), 2.01 (m, 1H), 1.24 (s, 3H), 1.08 (s, 3H), 1.06 (s, 3H), 0.99 (d, *J* = 6.9 Hz, 3H), 0.61 (d, *J* = 6.8 Hz, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 179.7, 166.7, 156.8, 140.3, 124.6, 117.8, 74.4, 72.9, 48.0, 42.2, 40.6, 40.2, 31.1, 27.4, 22.8, 21.3, 21.2, 19.9, 19.1; IR ν_{max} 2961, 1786, 1718, 1264, 1158, 1027, 982, 907 cm⁻¹; MS (EI, 70 eV) *m*/*z* 318 (M^{•+}, 5), 205 (5), 97 (100), 69 (15), 41 (25); HRMS *m*/*z* M^{•+} calcd for C₁₉H₂₆O₄ 318.1831, found 318.1832.

7-Hydroxy-6,8-dimethyl-3a,6,7,7a-tetrahydro-3,6-methanobenzofuran-2(3H)-one (20). A magnetically stirred solution of compound 18 (918 mg, 3.50 mmol) in methanol (55 mL) maintained at 0 °C was treated, in one portion, with potassium carbonate (242 mg, 1.75 mmol). The ensuing mixture was stirred at 0 °C for a further 0.5 h and then warmed to 22 °C, stirred at this temperature for 1 h, and then recooled to 0 °C and treated with water (30 mL) before being concentrated under reduced pressure. The residue thus obtained was extracted with ethyl acetate $(3 \times 100 \text{ mL})$, and the combined organic phases were then dried (MgSO₄), filtered, and concentrated under reduced pressure. The resulting light-yellow oil was subjected to flash column chromatography (silica, $3:7 \rightarrow 1:1 \text{ v/v}$ ethyl acetate/40–60 petroleum ether gradient elution) to give, after concentration of the appropriate fractions ($R_f = 0.1$ in 3:7 v/v ethyl acetate/40-60 petroleum ether), compound 20 (503 mg, 74%) as a white, crystalline solid: mp = 114–116 °C; $[\alpha]_D$ = +67.5 (*c* 1.0, CHCl₃); ¹H NMR (400 MHz, $CDCl_3$) δ 6.06–6.00 (complex m, 2H), 4.39 (t, J = 6.0 Hz, 1H), 3.43 (m, 1H), 3.26 (t, J = 6.7 Hz, 1H), 2.23 (d, J = 6.8 Hz, 1H), 2.11 (m, 1H), 2.01 (d, J = 4.5 Hz, 1H), 1.25 (s, 3H), 0.89 (d, J = 7.1 Hz, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 178.9, 141.2, 124.2, 74.9, 71.9, 46.9, 44.1, 40.4, 36.5, 18.8, 17.3; IR $\nu_{\rm max}$ 3417, 2968, 1760, 1750, 1184, 1102, 1015, 713 cm⁻¹; MS (ESI, + ve) m/z 217 [(M + Na)⁺, 100]; HRMS m/z (M + Na)⁺ calcd for C₁₁H₁₄O₃Na 217.0841, found 217.0844.

7-Hydroxy-8-isopropyl-6-methyl-3a,6,7,7a-tetrahydro-3,6-methanobenzofuran-2(3H)-one (21). A magnetically stirred solution of compound 19 (1.11 g, 3.50 mmol) in methanol (55 mL) maintained at 0 °C was treated, in one portion, with potassium carbonate (242 mg, 1.75 mmol). After being kept for 0.5 h at 0 °C, the reaction mixture was warmed to 22 °C, maintained at this temperature for 1 h, and then recooled to 0 °C and treated with water (30 mL) before being concentrated under reduced pressure. The residue thus obtained was extracted with ethyl acetate $(3 \times 100 \text{ mL})$, and the combined organic phases were then dried (MgSO₄), filtered, and concentrated under reduced pressure to give a light-yellow oil. This material was subjected to flash column chromatography (silica, 3:17 \rightarrow 3:7 v/v ethyl acetate/40-60 petroleum ether gradient elution) and gave, after concentration of the appropriate fractions ($R_f = 0.3$ in 3:7 v/v ethyl acetate/40-60 petroleum ether), compound 21 (590 mg, 76%) as a white solid: mp = 120–122 °C; $[\alpha]_D$ = +100.7 (c 1.0, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ 6.09 (d, J = 8.1 Hz, 1H), 5.93 (m, 1H), 4.35 (m, 1H), 3.43 (m, 1H), 3.12 (d, J = 6.9 Hz, 1H), 2.28 (m, 1H), 2.03-1.95 (complex m, 2H), 1.26 (s, 3H), 0.97 (d, J = 6.7 Hz, 3H),

0.60 (d, J = 6.7 Hz, 3H) (signal due to O–H group proton not observed); ¹³C NMR (100 MHz, CDCl₃) δ 179.3, 141.8, 123.3, 74.8, 72.9, 47.6, 44.0, 40.2 (3), 40.1 (7), 27.3, 22.7, 19.8, 18.8; IR ν_{max} 3425, 2958, 1765, 1170, 1100, 978, 723 cm⁻¹; MS (ESI, + ve) m/z 467 [(2M + Na)⁺, 30], 245 [(M + Na)⁺, 100], 223 [(M + H)⁺, 70]; HRMS m/z (M + Na)⁺ calcd for C₁₃H₁₈O₃Na 245.1154, found 245.1156.

6,8-Dimethyl-3a,7a-dihydro-3,6-methanobenzofuran-2,7-(3H,6H)-dione (22). A magnetically stirred solution of compound 20 (466 mg, 2.40 mmol) in dry dichloromethane (78 mL) maintained at 0 °C under an atmosphere of nitrogen was treated with pyridine (1.2 mL, 15.05 mmol) and then Dess-Martin periodinane (1.83 g, 4.32 mmol). After a further 0.5 h, the reaction mixture was warmed to 22 °C, maintained at this temperature for 3 h, and then poured into water (100 mL) and extracted with dichloromethane $(3 \times 100 \text{ mL})$. The combined organic phases were washed with NaOH (1 \times 100 mL of a 1 M aqueous solution) and then HCl $(1 \times 100 \text{ mL of a 1 M aqueous})$ solution) before being dried (MgSO₄), filtered, and then concentrated under reduced pressure. The residue thus obtained was subjected to flash column chromatography (silica, $1:9 \rightarrow 3:17$ v/v ethyl acetate/ 40-60 petroleum ether gradient elution) to give, after concentration of the appropriate fractions ($R_f = 0.3$ in 3:7 v/v ethyl acetate/40–60 petroleum ether), compound 22 (397 mg, 86%) as a white, crystalline solid: mp = 121–124 °C; $[\alpha]_D$ = +518.8 (c 1.0, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ 6.32 (dd, J = 8.1 and 6.5 Hz, 1H), 6.02 (d, J = 8.1 Hz, 1H), 4.15 (dd, J = 5.0 and 1.0 Hz, 1H), 3.66 (m, 1H), 2.37 (d, J = 5.0 Hz, 1H), 2.21 (m, 1H), 1.32 (s, 3H), 1.02 (d, J = 7.0 Hz, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 201.5, 177.8, 136.7, 128.3, 73.9, 53.8, 47.1, 41.3, 40.9, 16.1, 15.2; IR $\nu_{\rm max}$ 2976, 2937, 1785, 1736, 1159, 989, 837, 713 cm⁻¹; MS (ESI, + ve) m/z 247 [(M + MeOH + Na)⁺, 100], 215 [(M + Na)⁺, 20]; HRMS m/z (M + MeOH + Na)⁺ calcd for C12H16O4Na 247.0946, found 247.0946.

8-Isopropyl-6-methyl-3a,7a-dihydro-3,6-methanobenzofuran-2,7(3H,6H)-dione (23). A magnetically stirred solution of compound 21 (533 mg, 2.40 mmol) in dry dichloromethane (78 mL) maintained at 0 °C under an atmosphere of nitrogen was treated with pyridine (1.2 mL, 15.05 mmol) and then the Dess-Martin periodinane (1.83 g, 4.32 mmol). After a further 0.5 h the reaction mixture was warmed to 22 °C, stirred at this temperature for 3 h, and then poured into water (100 mL) and extracted with dichloromethane (3 \times 100 mL). The combined organic phases were washed with NaOH (1×100 mL of a 1 M aqueous solution) then HCl $(1 \times 100 \text{ mL of a 1 M aqueous})$ solution) before being dried (MgSO₄), filtered, and concentrated under reduced pressure. The residue thus obtained was subjected to flash column chromatography (silica, 3:17 \rightarrow 3:7 v/v ethyl acetate/ 40-60 petroleum ether gradient elution) and gave, after concentration of the appropriate fractions ($R_f = 0.3$ in 3:7 v/v ethyl acetate/40–60 petroleum ether), compound 23 (433 mg, 82%) as a white, crystalline solid: mp = 114–116 °C; $[\alpha]_D$ = +396.1 (c 0.8, CHCl₃); ¹H NMR $(400 \text{ MHz}, \text{CDCl}_3) \delta 6.19 \text{ (dd}, J = 8.1 \text{ and } 6.6 \text{ Hz}, 1\text{H}), 6.04 \text{ (m, 1H)},$ 4.12 (dd, J = 5.0 and 1.1 Hz, 1H), 3.67 (m, 1H), 2.64 (m, 1H), 2.11 (m, 1H), 1.99 (m, 1H), 1.36 (s, 3H), 1.01 (d, J = 6.9 Hz, 3H), 0.72 (d, J = 6.8 Hz, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 200.7, 178.5, 136.8, 127.3, 73.9, 53.6, 51.7, 40.9, 40.4, 28.2, 22.8, 20.0, 15.3; IR $\nu_{\rm max}$ 2961, 1802, 1786, 1734, 1151, 1002, 981, 899 cm⁻¹; MS (ESI, + ve) m/z 275 $[(M + MeOH + Na)^{+}, 100], 243 [(M + Na)^{+}, 40]; HRMS m/z (M + Na)^{+}, 100]$ MeOH + Na)⁺ calcd for $C_{14}H_{20}O_3Na$ 275.1259, found 275.1251.

7-Hydroxy-3a,8-dimethyl-3a,6,7,7a-tetrahydro-3,6-methanobenzofuran-2(3H)-one (24). A magnetically stirred solution of compound 16 (918 mg, 3.50 mmol) in methanol (55 mL) maintained at 0 °C under an atmosphere of nitrogen was treated, in one portion, with potassium carbonate (242 mg, 1.75 mmol). After a further 0.5 h, the reaction mixture was warmed to 22 °C, stirred at this temperature for 1 h, and then recooled to 0 °C and treated with water (30 mL) before being concentrated under reduced pressure. The residue thus obtained was extracted with ethyl acetate (3 × 100 mL), and the combined organic phases were then dried (MgSO₄), filtered, and concentrated under reduced pressure to give a light-yellow oil. This material was subjected to flash column chromatography (silica, 1:4 → 3:7 v/v ethyl acetate/40–60 petroleum ether gradient elution) and gave, after concentration of the appropriate fractions ($R_f = 0.2$ in 3:7 v/v ethyl

acetate/40–60 petroleum ether), compound **24** (538 mg, 79%) as a white, crystalline solid: mp = 146–148 °C; $[\alpha]_D$ = +15.8 (*c* 1.0, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ 6.36 (t, *J* = 7.6 Hz, 1H), 5.81 (dd, *J* = 7.6 and 1.0 Hz, 1H), 3.98 (d, *J* = 6.9 Hz, 1H), 3.57 (dd, *J* = 6.9 and 2.2 Hz, 1H), 2.68 (m, 1H), 2.47 (m, 2H), 1.68 (s, 1H), 1.39 (s, 3H), 0.93 (d, *J* = 7.1 Hz, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 179.0, 135.4, 130.8, 79.0, 69.0, 52.2, 45.2, 45.0, 33.1, 21.1, 20.1; IR ν_{max} 3402, 2964, 1740, 1347, 1182, 1085, 992, 948, 713, 703 cm⁻¹; MS (ESI, + ve) *m*/*z* 217 [(M + Na)⁺, 100%]; HRMS *m*/*z* (M + Na)⁺ calcd for C₁₁H₁₄O₃Na 217.0841, found 217.0842.

7-Hydroxy-8-isopropyl-3a-methyl-3a,6,7,7a-tetrahydro-3,6methanobenzofuran-2(3H)-one (25). A magnetically stirred solution of compound 17 (1.11 g, 3.50 mmol) in methanol (55 mL) maintained at 0 °C under an atmosphere of nitrogen was treated, in one portion, with potassium carbonate (242 mg, 1.75 mmol). After a further 0.5 h at 0 °C, the reaction mixture was warmed to 22 °C, stirred at this temperature for 1 h, and then recooled to 0 °C and treated with water (30 mL) before being concentrated under reduced pressure. The ensuing mixture was extracted with ethyl acetate $(3 \times$ 100 mL), and the combined organic phases were dried (MgSO₄), filtered, and concentrated under reduced pressure. The residue thus obtained was subjected to flash column chromatography (silica, $3:7 \rightarrow$ 1:1 v/v ethyl acetate/40-60 petroleum ether gradient elution) and gave, after concentration of the appropriate fractions ($R_f = 0.2$ in 3:7 v/v ethyl acetate/40-60 petroleum ether), compound 25 (732 mg, 94%) as a white, crystalline solid: mp = 60–61 °C; $[\alpha]_D$ = +2.8 (c 0.9, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ 6.35 (t, I = 8.3 Hz, 1H), 5.78 (dd, J = 8.3 and 1.2 Hz, 1H), 3.99 (d, J = 6.9 Hz, 1H), 3.54 (dd, J = 6.9 and 2.3 Hz, 1H), 2.89 (m, 1H), 2.48 (broad s, 1H), 1.85 (m, 2H), 1.38 (s, 3H), 1.25 (m, 1H), 0.93 (d, J = 6.5 Hz, 3H), 0.87 (d, J = 6.6 Hz, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 179.1, 134.6, 131.1, 79.3, 68.9, 49.7, 46.4, 44.9, 41.7, 30.7, 21.1, 20.8(0), 20.7(6); IR $\nu_{\rm max}$ 3452, 2962, 1778, 1166, 1117, 993, 706 cm⁻¹; MS (ESI, + ve) m/z 245 [(M + Na)⁺, 40], 223 $[(M + H)^+$, 100], 205 (50), 159 (100), 109 (90), 90 (70); HRMS m/z (M + H)⁺ calcd for C₁₃H₁₉O₃ 223.1334, found 223.1331.

3a,8-Dimethyl-3a,7a-dihydro-3,6-methanobenzofuran-2,7-(3H,6H)-dione (26). A magnetically stirred solution of compound 24 (466 mg, 2.40 mmol) in dry dichloromethane (78 mL) maintained at 0 °C under an atmosphere of nitrogen was treated with pyridine (1.2 mL, 15.05 mmol) and then the Dess-Martin periodinane (1.83 g, 4.32 mmol). After being kept at 0 °C for a further 0.5 h, the reaction mixture was warmed to 22 °C and maintained at this temperature for 3 h before being poured into water (100 mL) and extracted with dichloromethane $(3 \times 100 \text{ mL})$. The combined organic phases were washed with NaOH $(1 \times 100 \text{ mL of a 1 M aqueous solution})$ and HCl $(1 \times 100 \text{ mL of a 1 M aqueous solution})$ and then dried (MgSO₄), filtered, and concentrated under reduced pressure. The residue thus obtained was subjected to flash column chromatography (silica, 1:9 \rightarrow 3:17 v/v ethyl acetate/40-60 petroleum ether gradient elution) and gave, after concentration of the appropriate fractions ($R_f = 0.3$ in 3:7 v/v ethyl acetate/40-60 petroleum ether), compound 26 (434 mg, 94%) as a white, crystalline solid: mp = 135–136 °C; $[\alpha]_D = -445.7$ (c 1.0, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ 6.37 (t, J = 7.5 Hz, 1H), 6.13 (dd, J = 7.5 and 1.0 Hz, 1H), 3.75 (s, 1H), 3.27 (dd, J = 6.7 and 3.1 Hz, 1H), 2.59 (m, 1H), 1.98 (s, 1H), 1.47 (s, 3H), 1.02 (d, J = 7.0 Hz, 3H); 13 C NMR (100 MHz, CDCl₃) δ 200.0, 177.7, 134.5, 130.4, 78.5, 54.0, 51.9, 46.5, 37.2, 19.9, 18.4; IR ν_{max} 2964, 1788, 1742, 1312, 1161, 995, 875, 723 cm⁻¹; MS (ESI, + ve) m/z 247 [(M + MeOH + Na)⁺, 100]; HRMS m/z (M + MeOH + Na)⁺ Calcd for C₁₂H₁₆O₃Na 247.0946, found 247.0943.

8-Isopropyl-3a-methyl-3a,7a-dihydro-3,6-methanobenzofuran-2,7(3H,6H)-dione (27). A magnetically stirred solution of compound 25 (533 mg, 2.40 mmol) in dry dichloromethane (78 mL) maintained at 0 °C under an atmosphere of nitrogen was treated with pyridine (1.2 mL, 15.05 mmol) and then the Dess–Martin periodinane (1.83 g, 4.32 mmol). After a further 0.5 h at 0 °C, the reaction mixture was warmed to 22 °C and then stirred at this temperature for 3 h before being poured into water (100 mL) and extracted with dichloromethane (3 × 100 mL). The combined organic phases were washed with NaOH (1 \times 100 mL of a 1 M aqueous solution) and HCl (1 \times 100 mL of a 1 M aqueous solution) then dried (MgSO₄), filtered, and concentrated under reduced pressure. The residue thus obtained was subjected to flash column chromatography (silica, 1:9 \rightarrow 3:17 v/v ethyl acetate/40-60 petroleum ether gradient elution) and gave, after concentration of the appropriate fractions ($R_f = 0.5$ in 3:7 v/v ethyl acetate/40-60 petroleum ether), compound 27 (413 mg, 78%) as a white, crystalline solid: mp = 121-123 °C; $[\alpha]_{\rm D} = -332.5$ (c 1.0, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ 6.37 (t, J = 8.2 Hz, 1H), 6.09 (dd, J = 8.2 and 1.1 Hz, 1H), 3.77 (s, 1H), 3.49 (dd, J = 6.7 and 3.1 Hz, 1H), 2.19 (s, 1H), 1.93 (dd, J = 10.2 and 2.5 Hz, 1H), 1.47 (s, 3H), 1.42 (m, 1H), 1.00 (d, *J* = 6.5 Hz, 3H), 0.89 (d, *J* = 6.6 Hz, 3H); $^{13}\mathrm{C}$ NMR (100 MHz, CDCl₃) δ 200.4, 177.7, 134.8, 129.8, 78.8, 51.1, 50.2, 49.4, 46.2, 29.2, 20.6, 20.4, 19.9; IR $\nu_{\rm max}$ 2968, 17909, 1775, 1741, 1160, 985, 713 cm⁻¹; MS (ESI, + ve) m/z 275 [(M + MeOH + Na)⁺, 100], 221 [(M + H)⁺, 10]; HRMS m/z (M + MeOH + Na)⁺ calcd for C14H20O3Na 275.1259, found 275.1260.

(1aS,1a'R,3aS,4S,4aR,4a'R)-4,4a-Dimethylhexahydro-1Hcyclopropa[3,4]pentaleno[1,6-bc]furan-1,3(1aH)-dione (28). A deoxygenated and magnetically stirred solution of compound 22 (108 mg, 0.56 mmol) and acetophenone (90 μ L, 0.77 mmol) in acetone (100 mL) maintained under a nitrogen atmosphere was subjected to irradiation with a Hanovia 450 W medium-pressure mercury-vapor lamp for 0.83 h. The reaction mixture was then cooled and concentrated under reduced pressure to give a brown oil that was subjected to flash column chromatography (silica, $3:17 \rightarrow 3:7 \text{ v/v}$ ethyl acetate/40-60 petroleum ether gradient elution). Concentration of the appropriate fractions ($R_f = 0.3$ in 1:1 v/v ethyl acetate/40–60 petroleum ether) then gave compound 28 (44 mg, 40%) as a paleyellow oil: $[\alpha]_{D} = +5.1$ (c 1.0, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ 4.57 (dd, J = 9.0 and 2.2 Hz, 1H), 3.59 (m, 1H), 2.80 (t, J = 8.0 Hz, 1H), 2.60 (m, 1H), 2.45 (m, 1H), 1.98 (m, 1H), 1.30 (d, J = 7.2 Hz, 3H), 1.22 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 206.9, 176.8, 82.4, 57.2, 46.1, 45.1, 45.0, 44.7, 38.1, 19.5, 18.4; IR $\nu_{\rm max}$ 2969, 2928, 1779, 1726, 1192, 1152, 1054, 1031, 983 cm⁻¹; MS (ESI, + ve) m/z 215 [(M + Na)⁺, 100]; HRMS m/z (M + Na)⁺ calcd for C₁₁H₁₂O₃Na 215.0684, found 215.0687.

(1aS, 1a' R, 3aS, 4S, 4aS, 4a' R)-4-Isopropyl-4a-methylhexahydro-1Hcyclopropa[3,4]pentaleno[1,6-bc]furan-1,3(1aH)-dione (29). A deoxygenated and magnetically stirred solution of compound 23 (123 mg, 0.56 mmol) and acetophenone (90 μ L, 0.77 mmol) in acetone (100 mL) maintained under a nitrogen atmosphere was subjected to irradiation with a Hanovia 450 W medium-pressure mercury-vapor lamp for 0.67 h. The reaction mixture thus formed was cooled and concentrated under reduced pressure to give a yellow oil that was subjected to flash column chromatography (silica, $3:17 \rightarrow 1:3 \text{ v/v}$ ethyl acetate/40-60 petroleum ether gradient elution). Concentration of the appropriate fractions ($R_f = 0.4$ in 1:1 v/v ethyl acetate/40–60 petroleum ether) then gave compound 29 (53 mg, 43%) as a white, crystalline solid: mp = 107–109 °C; $[\alpha]_{\rm D} = -7.6$ (c 1.0, CHCl₃). ¹H NMR (400 MHz, CDCl₃) δ 4.60 (dd, J = 8.9 and 2.0 Hz, 1H), 3.61 (m, 1H), 3.10 (dd, J = 8.5 and 6.5 Hz, 1H), 2.62 (m, 1H), 2.38 (m, 1H), 2.15 (m, 1H), 1.88 (m, 1H), 1.27 (s, 3H), 1.05 (d, J = 6.9 Hz, 3H), 0.97 (d, J = 6.5 Hz, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 206.9, 177.4, 82.5, 55.0, 47.8, 46.6, 45.6, 44.0, 38.3, 29.2, 22.1, 19.5, 17.0; IR $\nu_{\rm max}$ 2963, 1780, 1728, 1196, 1150, 1049, 983 cm $^{-1};$ MS (EI, 70 eV) m/z 220 (M^{•+}, 25), 153 (40), 149 (90), 133 (40), 105 (60), 93 (100), 91 (70), 84 (60), 77 (50), 59 (70); HRMS m/z M^{•+} calcd for C13H16O3 220.1099, found 220.1092.

(2aS,2a'R,3aR,6S,6aS)-5,6-Dimethyl-2a¹,3a,6,6a-tetrahydro-1Hcyclobuta[cd]isobenzofuran-1,3(2aH)-dione (**30**). A deoxygenated and magnetically stirred solution of compound **22** (100 mg, 0.52 mmol) in dichloromethane (100 mL) maintained under a nitrogen atmosphere was subjected to irradiation with a Hanovia 450 W medium-pressure mercury-vapor lamp for 0.5 h. The reaction thus obtained was then cooled and concentrated under reduced pressure to give a yellow solid. Subjection of this material to flash column chromatography (silica, 1:4 \rightarrow 3:7 v/v ethyl acetate/40–60 petroleum ether gradient elution) gave, after concentration of the appropriate fractions ($R_f = 0.2$ in 3:7 v/v ethyl acetate/40–60 petroleum ether), compound **30** (75 mg, 75%) as a white, crystalline solid: mp = 161– 163 °C; $[\alpha]_{\rm D}$ = -344.2 (*c* 1.0, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ 5.37 (m, 1H), 5.33 (m, 1H), 3.93 (m, 1H), 3.48 (m, 1H), 2.97 (dd, *J* = 10.0 and 2.1 Hz, 1H), 2.67 (m, 1H), 1.78 (s, 3H), 1.19 (d, *J* = 7.2 Hz, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 200.3, 177.5, 140.6, 112.9, 89.8, 58.8, 43.3, 33.9, 26.8, 23.0, 19.8; IR $\nu_{\rm max}$ 2968, 1782, 1256, 1148, 1141, 1130, 1004 cm⁻¹; MS (ESI, + ve) *m*/*z* 247 [(M + MeOH + Na)⁺, 100]; HRMS *m*/*z* (M + MeOH + Na)⁺ calcd for C₁₂H₁₆O₃Na 247.0946, found 247.0945.

(2aS,2a'R,3aR,6S,6aS)-6-Isopropyl-5-methyl-2a',3a,6,6a-tetrahydro-1H-cyclobuta[cd]isobenzofuran-1,3(2aH)-dione (31). A deoxygenated and magnetically stirred solution of compound 23 (100 mg, 0.45 mmol) in dichloromethane (100 mL) maintained under nitrogen was subjected to irradiation with a Hanovia 450 W medium-pressure mercury-vapor lamp for 0.5 h. The reaction mixture thus obtained was cooled and concentrated under reduced pressure to give a yellow oil. Subjection of this material to flash column chromatography (silica, 1:4 \rightarrow 3:7 v/v ethyl acetate/40–60 petroleum ether gradient elution) gave, after concentration of the appropriate fractions ($R_f = 0.2$ in 3:7 v/v ethyl acetate/40-60 petroleum ether), compound 31 (73 mg, 73%) as a clear, colorless oil: $[\alpha]_D = -163.3$ (c 0.7, CHCl₃). ¹H NMR (400 MHz, CDCl₃) δ 5.47 (s, 1H), 5.27 (dd, J = 6.4 and 4.0 Hz, 1H), 3.88 (m, 1H), 3.43 (m, 1H), 3.08 (dd, J = 10.1 and 1.4 Hz, 1H), 2.45 (d, J = 3.9 Hz, 1 H), 1.90 (m, 1H), 1.79 (s, 3H), 1.06 (d, J = 6.8 Hz,3H), 0.89 (d, J = 6.9 Hz, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 200.4, 178.7, 138.9, 114.7, 89.1, 59.4, 45.9, 38.1, 30.8, 28.3, 24.5, 21.6, 19.6; IR ν_{max} 2963, 1782, 1168, 1147, 1131, 1005, 838 cm⁻¹; MS (ESI, + ve) m/z 275 [(M + MeOH + Na)⁺, 100], 221 [(M + H)⁺, 25]; HRMS m/ $z (M + MeOH + Na)^+$ calcd for $C_{14}H_{20}O_3Na$ 275.1259, found 275.1260.

(2aS, 2a'R, 3S, 5aR, 5bR)-3, 4-Dimethyl-2a', 3, 5a, 5btetrahydrocyclopropa[cd]isobenzofuran-2(2aH)-one (32). A deoxygenated and magnetically stirred solution of compound 22 (100 mg, 0.52 mmol) in dichloromethane (100 mL) maintained under a nitrogen atmosphere was irradiated with a Hanovia 450 W mediumpressure mercury-vapor lamp for 5 h. The ensuing reaction mixture was cooled to ca. 22 °C then concentrated under reduced pressure to give a brown oil. Subjection of this material to flash column chromatography (silica, 1:9 \rightarrow 3:7 v/v ethyl acetate/40–60 petroleum ether gradient elution) gave two fractions, A and B.

Concentration of fraction A ($R_f = 0.5$ in 3:7 v/v ethyl acetate/40– 60 petroleum ether) afforded compound **32** (32 mg, 37%) as a colorless, crystalline solid: mp = 73–75 °C; $[\alpha]_D = -102.8$ (*c* 0.4, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ 5.34 (s, 1H), 4.33 (dd, *J* = 6.3 and 5.4 Hz, 1H), 2.89 (dd, *J* = 7.7 and 3.0 Hz, 1H), 2.23 (m, 1H), 1.95 (m, 1H), 1.69 (s, 3H), 1.46 (m, 1H), 1.21 (d, *J* = 7.0 Hz, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 180.3, 142.7, 112.7, 60.5, 44.7, 35.9, 22.4, 18.7, 17.8, 11.6; IR ν_{max} 2965, 1773, 1453, 1150, 1086, 1074, 922, 794 cm⁻¹; MS (ESI, + ve) *m/z* 187 [(M + Na)⁺, 100]; HRMS *m/z* (M + Na)⁺ calcd for C₁₀H₁₂O₂Na 187.0735, found 187.0735.

Concentration of fraction B ($R_f = 0.2$ in 3:7 v/v ethyl acetate/40–60 petroleum ether) afforded compound **30** (28 mg, 28%) as a white solid that was identical with an authentic sample.

(2aS, 2a'R, 3S, 5aR, 5bR)-3-IsopropyI-4-methyI-2a', 3, 5a, 5btetrahydrocyclopropa[cd]isobenzofuran-2(2aH)-one (33). A deoxygenated and magnetically stirred solution of compound 23 (100 mg, 0.45 mmol) in dichloromethane (100 mL) maintained under a nitrogen atmosphere was subjected to irradiation with a Hanovia 450 W medium-pressure mercury-vapor lamp for 1.5 h. The ensuing mixture was cooled then concentrated under reduced pressure to give a brown oil. Subjection of this material to flash column chromatography (silica, 1:9 \rightarrow 3:7 v/v ethyl acetate/40–60 petroleum ether gradient elution) afforded two fractions, A and B.

Concentration of fraction A ($R_f = 0.6$ in 3:7 v/v ethyl acetate/40– 60 petroleum ether) gave compound **33** (45 mg, 52%) as a clear, colorless oil: [α]_D = -29.8 (*c* 1.0, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ 5.45 (*s*, 1H), 4.33 (m, 1H), 3.03 (dd, *J* = 7.7 and 2.4 Hz, 1H), 2.02–1.88 (complex m, 3H), 1.71 (*s*, 3H), 1.45 (m, 1H), 1.06 (d, *J* = 6.5 Hz, 3H), 0.98 (d, *J* = 6.6 Hz, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 181.1, 141.1, 114.8, 61.4, 47.6, 40.3, 30.7, 24.2, 21.7, 19.9, 18.0, 13.5; IR $\nu_{\rm max}$ 2960, 1769, 1149, 1127, 1087, 933, 781 cm⁻¹; MS (ESI, + ve) m/z 215 [(M + Na)⁺, 100]; HRMS m/z (M + Na)⁺ calcd for C₁₂H₁₆O₂Na 215.1048, found 215.1048.

Concentration of fraction B ($R_f = 0.2$ in 3:7 v/v ethyl acetate/40–60 petroleum ether) afforded compound **31** (25 mg, 25%) as a white solid that was identical with an authentic sample.

(1aR,1a'S,3aR,4S,4a¹R)-1a',4-Dimethylhexahydro-1Hcyclopropa[3,4]pentaleno[1,6-bc]furan-1,3(1aH)-dione (34). A deoxygenated and magnetically stirred solution of compound 26 (108 mg, 0.56 mmol) and acetophenone (90 μ L, 0.77 mmol) in acetone (100 mL) maintained under nitrogen was subjected to irradiation with a Hanovia 450 W medium-pressure mercury-vapor lamp for 2.5 h. The ensuing reaction mixture was cooled then concentrated under reduced pressure to give a yellow oil. Subjection of this material to flash column chromatography (silica, $3:17 \rightarrow 3:7 \text{ v/v}$ ethyl acetate/40-60 petroleum ether gradient elution) gave, after concentration of the appropriate fractions ($R_f = 0.3$ in 1:1 v/v ethyl acetate/40-60 petroleum ether), compound 34 (49 mg, 45%) as a white, crystalline solid: mp = $154-156 \,^{\circ}C_{1} [\alpha]_{D} = +68.6 (c \ 1.0, CHCl_{3})$. ¹H NMR (400 MHz, CDCl₃) δ 4.21 (s, 1H), 2.55–2.47 (complex m, 3H), 2.18 (m, 1H), 1.97 (m, 1H), 1.54 (s, 3H), 1.38 (d, J = 6.8 Hz, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 206.9, 176.5, 87.7, 64.3, 55.0, 42.6, 40.3, 38.0, 35.6, 22.6, 20.8; IR ν_{max} 2970, 1781, 1726, 1238, 1163, 1033, 884, 874 cm^{-1} ; MS (ESI, + ve) m/z 247 [(M + MeOH + Na)⁺, 20], 215 [(M + Na)⁺, 100], 193 (30); HRMS m/z (M + Na)⁺ calcd for C₁₁H₁₂O₃Na 215.0684, found 215.0676.

(1aR,1a'S,3aR,4S,4a'R)-4-IsopropyI-1a'-methylhexahydro-1Hcyclopropa[3,4]pentaleno[1,6-bc]furan-1,3(1aH)-dione (35). A deoxygenated and magnetically stirred solution of compound 27 (123 mg, 0.56 mmol) and acetophenone (90 μ L, 0.77 mmol) in acetone (100 mL) maintained under a nitrogen atmosphere was irradiated with a Hanovia 450 W medium-pressure mercury-vapor lamp for 2.5 h. The cooled reaction mixture was concentrated under reduced pressure to give a yellow oil and this subjected to flash column chromatography (silica, $3:17 \rightarrow 1:3 \text{ v/v}$ ethyl acetate/40–60 petroleum ether gradient elution). Concentration of the appropriate fractions ($R_f = 0.5$ in 1:1 v/ v ethyl acetate/40-60 petroleum ether) then gave compound 35 (60 mg, 49%) as a white, crystalline solid: mp = 102–103 °C; $[\alpha]_D$ = +83.8 (c 1.0, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ 4.22 (s, 1H), 2.72 (d, J = 6.4 Hz, 1H), 2.41 (t, J = 4.0 Hz, 1H), 2.22–2.15 (complex m, 2H), 2.08 (m, 1H), 1.94 (m, 1H), 1.53 (s, 3H), 1.03 (d, J = 3.0 Hz, 3H), 1.02 (d, J = 3.0 Hz, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 207.1, 176.8, 87.6, 59.1, 54.6, 53.7, 37.4, 37.3, 35.5, 32.9, 20.6, 20.2, 19.4; IR $\nu_{\rm max}$ 2961, 1790, 1781, 1732, 1240, 1160, 1047, 889 ${\rm cm^{-1}};$ MS (ESI, + ve) m/z 275 [(M + MeOH + Na)⁺, 20], 243 [(M + Na)⁺, 100]; HRMS m/z (M + Na)⁺ calcd for C₁₃H₁₆O₃Na 243.0997, found 243.0993

(2aR,2a'S,3aS,6S,6aR)-2a',6-Dimethyl-2a',3a,6,6a-tetrahydro-1H-cyclobuta[cd]isobenzofuran-1,3(2aH)-dione (36). A deoxygenated and magnetically stirred solution of compound 26 (100 mg, 0.52 mmol) in dichloromethane (100 mL) maintained under a nitrogen atmosphere was irradiated with a Hanovia 450 W medium-pressure mercury-vapor lamp for 0.58 h then cooled and concentrated under reduced pressure to give a yellow solid. Subjection of this material to flash column chromatography (silica, 1:4 \rightarrow 3:7 v/v ethyl acetate/40– 60 petroleum ether gradient elution) then gave, after concentration of the appropriate fractions ($R_f = 0.2$ in 3:7 v/v ethyl acetate/40-60 petroleum ether), compound 36 (70 mg, 70%) as a white, crystalline solid: mp = 100–101 °C; $[\alpha]_D$ = +220.5 (c 1.0, CHCl₃). ¹H NMR (400 MHz, CDCl₃) δ 6.02 (m, 1H), 5.69 (dd, J = 10.0 and 4.0 Hz, 1H), 4.82 (d, J = 4.0 Hz, 1H), 3.60 (m, 1H), 2.93 (m, 1H), 2.66 (s, 1H), 1.70 (s, 3H), 1.23 (d, J = 7.3 Hz, 3H); ¹³C NMR (100 MHz, $CDCl_3$) δ 200.1, 177.7, 132.6, 119.1, 93.5, 63.8, 48.5, 35.5, 30.1, 24.9, 21.2; IR $\nu_{\rm max}$ 2965, 1779, 1191, 1096, 997, 744, 725 cm $^{-1}$; MS (ESI, + ve) m/z 247 [(M + MeOH + Na)⁺, 100%]; HRMS m/z (M + MeOH + Na)⁺ calcd for C₁₂H₁₆O₃Na 247.0946, found 247.0947.

(2aR,2a'S,3aS,6S,6aR)-6-Isopropyl-2a'-methyl-2a',3a,6,6a-tetrahydro-1H-cyclobuta[cd]isobenzofuran-1,3(2aH)-dione (**37**). A deoxygenated and magnetically stirred solution of compound **27** (100 mg, 0.45 mmol) in dichloromethane (100 mL) maintained under a

nitrogen atmosphere was irradiated with a Hanovia 450 W mediumpressure mercury-vapor lamp for 0.55 h then cooled and concentrated under reduced pressure to give a yellow solid. Subjection of this material to flash column chromatography (silica, $1:4 \rightarrow 3:7 \text{ v/v}$ ethyl acetate/40-60 petroleum ether gradient elution) then gave, after concentration of the appropriate fractions ($R_f = 0.2$ in 3:7 v/v ethyl acetate/40-60 petroleum ether), compound 37 (71 mg, 71%) as a white, crystalline solid: mp = 96–98 °C; $[\alpha]_D$ = +196.2 (c 1.0, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ 6.11 (m, 1H), 5.73 (dd, J = 10.1 and 3.6 Hz, 1H), 4.79 (d, J = 3.6 Hz, 1H), 3.56 (m, 1H), 2.95 (d, I = 1.7 Hz, 1H), 2.43 (t, I = 8.0 Hz, 1H), 1.68 (s, 3H), 1.66-1.60 (complex m, 1H), 1.09 (d, J = 6.6 Hz, 3H), 1.01 (d, J = 6.6 Hz, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 199.9, 178.5, 131.8, 120.0, 93.5, 64.2, 44.8, 42.6, 35.9, 33.1, 24.1, 21.4, 21.3; IR $\nu_{\rm max}$ 2964, 1782, 1471, 1293, 1193, 1000, 722 cm⁻¹; MS (ESI, + ve) m/z 527 [(2M + 2MeOH + Na)⁺, 15], 275 [(M + MeOH + Na)⁺, 100], 221 [(M + H)⁺, 15]; HRMS m/z (M + MeOH + Na)⁺ calcd for C₁₄H₂₀O₃Na 275.1259, found 275.1257.

 $(2aR, 2a'S, 3S, 5aS, 5bS) - 2a', 3-Dimethyl-2a^1, 3, 5a, 5b$ tetrahydrocyclopropa[cd]isobenzofuran-2(2aH)-one (**38**). A deoxygenated and magnetically stirred solution of compound**26**(100 mg,0.52 mmol) in dichloromethane (100 mL) maintained under anitrogen atmosphere was irradiated with a Hanovia 450 W mediumpressure mercury-vapor lamp for 3.1 h then cooled and concentratedunder reduced pressure to give a brown oil. Subjection of this material $to flash column chromatography (silica, 1:9 <math>\rightarrow$ 3:7 v/v ethyl acetate/ 40–60 petroleum ether gradient elution) then gave two fractions, A and B.

Concentration of fraction A ($R_f = 0.5$ in 3:7 v/v ethyl acetate/40– 60 petroleum ether) afforded compound **38** (30 mg, 35%) as a colorless, crystalline solid: mp = 40–42 °C; [α]_D = +152.0 (*c* 0.5, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ 5.95 (dd, *J* = 9.9 and 6.0 Hz, 1H), 5.66 (dd, *J* = 9.9 and 4.5 Hz, 1H), 4.03 (d, *J* = 5.6 Hz, 1H), 2.65 (s, 1H), 2.54 (m, 1H), 1.39 (s, 3H), 1.35 (m, 1H), 1.22 (d, *J* = 7.1 Hz, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 180.7, 134.8, 119.4, 65.2, 49.3, 31.7, 23.5, 20.6, 19.3, 18.9; IR ν_{max} 2962, 2929, 1783, 1769, 1143, 1090, 846, 750 cm⁻¹; MS (EI, 70 eV) *m*/*z* 164 (M^{+•}, 20%), 121 (30), 107 (100), 91 (70), 77 (25), 65 (15); HRMS *m*/*z* M^{•+} calcd for C₁₀H₁₂O₂ 164.0837, found 164.0835.

Concentration of fraction B ($R_f = 0.2$ in 3:7 v/v ethyl acetate/40–60 petroleum ether) afforded compound **36** (32 mg, 32%) as a white, crystalline solid that was identical, in all respects, with an authentic sample.

(2aR, 2a'S, 3S, 5aS, 5bS)-3-IsopropyI-2a'-methyI-2a', 3, 5a, 5btetrahydrocyclopropa[cd]isobenzofuran-2(2aH)-one (**39**). A deoxygenated and magnetically stirred solution of compound **27** (100 mg, 0.45 mmol) in dichloromethane (100 mL) maintained under a nitrogen atmosphere was irradiated with a Hanovia 450 W mediumpressure mercury-vapor lamp for 3.1 h then cooled and concentrated under reduced pressure to give a brown oil. Subjection of this material to flash column chromatography (silica, 1:9 \rightarrow 3:7 v/v ethyl acetate/ 40–60 petroleum ether elution) then gave two fractions, A and B.

Concentration of fraction A ($R_f = 0.6$ in 3:7 v/v ethyl acetate/40– 60 petroleum ether) afforded compound **39** (35 mg, 40%) as a yellow oil, [α]_D = +166.7 (c 1.0, CHCl₃). ¹H NMR (400 MHz, CDCl₃) δ 6.02 (dd, J = 9.9 and 6.1 Hz, 1H), 5.70 (dd, J = 9.9 and 4.2 Hz, 1H), 4.02 (d, J = 5.5 Hz, 1H), 2.92 (s, 1H), 2.02 (m, 1H), 1.79 (m, 1H), 1.36 (s, 3H), 1.34 (m, 1H), 1.05 (d, J = 6.6 Hz, 3H), 0.99 (d, J = 6.6 Hz, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 181.2, 134.0, 120.3, 65.5, 46.0, 43.8, 32.3, 23.7, 21.8, 20.9, 20.1, 19.3; IR ν_{max} 2966, 2872, 1778, 1148, 1092, 834, 752 cm⁻¹; MS (ESI, + ve) m/z 215 [(M + Na)⁺, 100%]; HRMS m/z (M + Na)⁺ calcd for C₁₂H₁₆O₂Na 215.1048, found 215.1047.

Concentration of fraction B ($R_f = 0.2$ in 3:7 v/v ethyl acetate/40–60 petroleum ether) afforded compound 37 (34 mg, 34%) as a white, crystalline solid that was identical, in all respects, with an authentic sample.

(3*aR*, 4*R*, 4*aR*, 5*S*, 7*aR*, 8*S*, 8*aS*) - 2, 2, 4, 6, 6 - Pentamethyl-3*a*, 4*a*, 5, 6, 7, 7*a*, 8, 8*a*-octahydro-4H-4, 8-ethenoindeno[5, 6-d][1, 3]dioxol-5-yl Acetate (42). A magnetically stirred solution of compound 41^{4d, 5b} (5.00 g, 17.96 mmol), acetic anhydride (3.40 mL, 35.97 mmol) and 4-(N,N-dimethylamino)pyridine (440 mg, 3.60 mmol) in triethylamine (7.50 mL, 53.8 mmol) was stirred at 0 °C for 2 h and then warmed to 22 °C and stirred at this temperature for another 2 h. After this time, NH₄Cl (50 mL of a saturated aqueous solution) was added to the reaction mixture, and the resulting suspension was extracted with dichloromethane $(3 \times 50 \text{ mL})$. The combined organic fractions were washed with water $(2 \times 50 \text{ mL})$ and brine $(1 \times 50 \text{ mL})$ before being dried (MgSO₄), filtered, and concentrated under reduced pressure. The residue thus obtained was subjected to flash column chromatography (silica, 1:19 v/v ethyl acetate/40-60 petroleum ether elution) and gave, after concentration of the appropriate fractions (R_{f} = 0.3 in 1:9 v/v ethyl acetate/40–60 petroleum ether), compound 42(5.58 g, 97%) as a white, crystalline solid: mp = 135–137 °C; $[\alpha]_D$ = +7.6 (c 1.0, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ 5.86–5.80 (complex m, 2H), 4.98 (d, J = 6.1 Hz, 1H), 4.22 (dd, J = 7.2 and 3.1 Hz, 1H), 3.79 (d, J = 7.2 Hz, 1H), 2.75 (m, 1H), 2.22 (m, 1H), 2.07 (m, 1H), 2.04 (s, 3H), 1.37 (m, 2H), 1.32 (s, 3H), 1.27 (s, 3H), 1.14 (s, 3H), 0.95 (s, 3H), 0.82 (s, 3H); ^{13}C NMR (100 MHz, CDCl₃) δ 170.4, 137.3, 125.7, 109.1, 84.1, 81.0, 80.1, 49.7, 44.7, 42.0, 40.8, 40.0, 38.0, 25.6 (5), 25.6 (2), 25.2, 22.4, 21.6, 19.9; IR $\nu_{\rm max}$ 2961, 2934, 1733, 1370, 1234, 1073, 723 cm⁻¹; MS (ESI, + ve) m/z 343 [(M + Na)⁺, 100]; HRMS m/z (M + Na)⁺ calcd for C₁₉H₂₈O₄Na 343.1885, found 343.1884.

(3S,3aR,4R,7S,7aR,8S,9R)-8,9-Dihydroxy-2,2,4-trimethyl-2,3,3a,4,7,7a-hexahydro-1H-4,7-ethanoinden-3-yl Acetate (43). AG-50W-X8 resin (3.32 g of H⁺ form) was added to a magnetically stirred solution of compound 42 in methanol/water (72 mL of a 5:1 v/v mixture) and the resulting suspension heated under reflux for 67 h and then cooled to 22 °C before being filtered through a sintered glass funnel. The filtrate was concentrated under reduced pressure and the residue thus obtained subjected to flash column chromatography (silica, 3:7 v/v ethyl acetate/40–60 petroleum ether elution). Concentration of the appropriate fractions ($R_f = 0.1$ in 3:7 v/v ethyl acetate/40-60 petroleum ether) gave compound 43 (1.80 g, 97%) as a white solid: mp = 73–74 °C; $[\alpha]_{\rm D}$ = -2.3 (*c* 1.1, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ 5.99–5.91 (complex m, 2H), 4.97 (d, J = 6.1 Hz, 1H), 3.88 (dd, J = 7.5 and 2.4 Hz, 1H), 3.40 (d, J = 7.5 Hz, 1H), 2.73 (m, 1H), 2.52 (broad s, 2H), 2.27 (m, 1H), 2.13 (m, 1H), 2.03 (s, 3H), 1.39 (m, 1H), 1.25 (t, J = 11.6 Hz, 1H), 1.15 (s, 3H), 0.93 (s, 3H), 0.80 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 170.4, 138.8, 126.8, 81.6, 75.4, 71.7, 50.4, 44.1, 42.0, 41.5, 41.3, 40.8, 25.5, 22.3, 21.6, 19.6; IR $\nu_{\rm max}$ 3394, 2965, 2932, 1732, 1374, 1240, 1057 cm $^{-1}$ MS m/z 583 [(2M + Na)⁺, 5], 303 [(M + Na)⁺, 100]; HRMS m/z (M + Na)⁺ calcd for $C_{16}H_{24}O_4Na$ 303.1572, found 303.1577.

(3S, 3aR, 4R, 7S, 7aS, 9R)-9-Hydroxy-2, 2, 4-trimethyl-8-oxo-2,3,3a,4,7,7a-hexahydro-1H-4,7-ethanoinden-3-yl Acetate (44). A magnetically stirred mixture of p-toluenesulfonic acid monohydrate (448 mg, 2.36 mmol) and 4-acetamido-TEMPO (502 mg, 2.35 mmol) in dichloromethane (10 mL) was maintained at 0 °C under a nitrogen atmosphere for 0.5 h and the resulting solution then added, dropwise over 2 h, to a magnetically stirred solution of compound 43 (300 mg, 1.07 mmol) in dichloromethane (10 mL) maintained at 0 °C. The ensuing mixture was stirred for a further 2 h at this temperature and then quenched with $NaHCO_3$ (50 mL of a saturated aqueous solution). The resulting suspension was extracted with dichloromethane $(3 \times 50 \text{ mL})$, and the combined organic phases were washed with water (2 \times 50 mL) and brine (1 \times 50 mL) before being dried (Na_2SO_4) , filtered, and concentrated under reduced pressure. The residue thus obtained was subjected to flash column chromatography (silica, 3:17 v/v ethyl acetate/40-60 petroleum ether elution) to give, after concentration of the appropriate fractions ($R_f = 0.2$ in 3:7 v/v ethyl acetate/40-60 petroleum ether), compound 44 (289 mg, 97%) as a white, crystalline solid: mp = 147–149 °C; $[\alpha]_D$ = +215.0 (c 1.1, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ 6.08 (d, *J* = 7.9 Hz, 1H), 5.92 (t, J = 7.9 Hz, 1H), 5.09 (d, J = 6.0 Hz, 1H), 3.34 (s, 1H), 3.15 (d, J = 1000 Hz)6.0 Hz, 1H), 2.72–2.49 (complex m, 3H), 2.06 (s, 3H), 1.52 (dd, J = 12.2 and 7.4 Hz, 1H), 1.40 (t, J = 12.2 Hz, 1H), 1.22 (s, 3H), 1.00 (s, 3H), 0.87 (s, 3H); $^{13}\mathrm{C}$ NMR (100 MHz, CDCl3) δ 211.0, 170.2, 142.2, 121.9, 81.1, 75.2, 51.8, 50.1, 44.8, 43.9, 41.7, 39.3, 25.6, 22.2, 21.6, 18.1; IR $\nu_{\rm max}$ 3451, 3239, 2965, 1730, 1374, 1236, 1081, 1039

cm⁻¹; MS (ESI, + ve) m/z 579 [(2M + Na)⁺, 10], 301 [(M + Na)⁺, 100]; HRMS m/z (M + Na)⁺ calcd for C₁₆H₂₂O₄Na 301.1416, found 301.1411.

(3S,3aR,4R,7S,7aS,9R)-2,2,4-Trimethyl-8-oxo-2,3,3a,4,7,7a-hexahydro-1H-4,7-ethanoindene-3,9-diyl Diacetate (45). A magnetically stirred solution of compound 44 (239 mg, 0.86 mmol), 4-(N,Ndimethylamino)pyridine (11 mg, 0.09 mmol), and triethylamine (180 μ L, 1.29 mmol) in dichloromethane (10 mL) maintained at 0 °C under a nitrogen atmosphere was treated with acetic anhydride (97 μ L, 1.03 mmol). The ensuing mixture was kept at 0 °C for 2 h and then at 22 °C for 16 h before being quenched with NH₄Cl (10 mL of a saturated aqueous solution). The resulting mixture was extracted with dichloromethane $(2 \times 20 \text{ mL})$, and the combined organic phases were washed with water $(2 \times 20 \text{ mL})$ and brine $(1 \times 20 \text{ mL})$ before being dried (Na₂SO₄), filtered, and concentrated under reduced pressure. The residue thus obtained was subjected to flash column chromatography (silica, 3:17 v/v ethyl acetate/40-60 petroleum ether elution) and gave, after concentration of the appropriate fractions ($R_f = 0.4$ in 3:7 v/v ethyl acetate/40–60 petroleum ether), compound 45 (250 mg, 91%) as a white, crystalline solid: mp = 175-177 °C; $[\alpha]_{\rm D}$ = +200.0 (c 1.0, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ 6.09 (d, J = 8.1 Hz, 1H), 5.97 (t, J = 8.1 Hz, 1H), 5.09 (d, J = 6.0 Hz, 1H), 4.85 (s, 1H), 3.17 (d, J = 6.0 Hz, 1H), 2.74 (m, 1H), 2.64 (dd, J = 10.6 and 6.0 Hz, 1H), 2.10 (s, 3H), 2.06 (s, 3H), 1.53 (dd, J = 12.2 and 7.3 Hz, 1H), 1.39 (t, J = 12.2 Hz, 1H), 1.05 (s, 3H), 1.01 (s, 3H), 0.86 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 205.2, 170.7, 170.1, 141.4, 122.2, 81.1, 74.2, 51.5, 50.7, 44.9, 42.6, 41.7, 39.6, 25.5, 22.2, 21.5, 20.7, 17.9; IR $\nu_{\rm max}$ 2979, 2965, 2932, 1721, 1374, 1230, 1053, 1020 cm⁻¹; MS (ESI, + ve) m/z 663 [(2M + Na)⁺, 20], 343 [(M + Na)⁺, 100], 321 [(M + H)⁺, 5]; HRMS m/z (M + Na)⁺ calcd for C₁₈H₂₄O₅Na 343.1521, found 343.1505.

(3S,3aR,4R,7S,7aS,9R)-9-(Methoxymethoxy)-2,2,4-trimethyl-8oxo-2.3.3a,4,7.7a-hexahvdro-1H-4,7-ethanoinden-3-vl Acetate (46). A magnetically stirred solution of compound 44 (280 mg, 1.00 mmol), chloromethyl methyl ether (122 µL, 1.60 mmol), tetra-nbutylammonium iodide (37 mg, 0.10 mmol), and N,N-diisopropylethylamine (348 μ L, 2.00 mmol) in dichloromethane (5 mL) maintained under a nitrogen atmosphere was stirred at 22 °C for 48 h and then treated with NH₄Cl (10 mL of a saturated aqueous solution) and extracted with dichloromethane (2 \times 20 mL). The combined organic phases were washed with water $(2 \times 20 \text{ mL})$ and brine $(1 \times 20 \text{ mL})$ before being dried (Na₂SO₄), filtered, and concentrated under reduced pressure. The residue thus obtained was subjected to flash column chromatography (silica, 1:4 v/v ethyl acetate/40-60 petroleum ether elution) and gave, after concentration of the appropriate fractions ($R_f = 0.3$ in 1:4 v/v ethyl acetate/40–60 petroleum ether), compound 46 (251 mg, 77%) as a pale-yellow oil: $[\alpha]_{\rm D}$ = +213.0 (c 1.0, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ 6.06 (d, J = 8.0 Hz, 1H), 5.90 (t, J = 8.0 Hz, 1H), 5.03 (d, J = 6.7 Hz, 2H), 4.59 (d, J = 6.7 Hz, 1H), 3.38 (s, 3H), 3.31 (s, 1H), 3.04 (d, J = 6.3 Hz, 1H), 2.61 (m, 1H), 2.46 (dd, J = 10.6 and 6.1 Hz, 1H), 2.03 (s, 3H), 1.46 (dd, J = 12.1 and 7.4 Hz, 1H), 1.35 (t, J = 12.1 Hz, 1H), 1.14 (s, 3H), 0.95 (s, 3H), 0.82 (s, 3H); $^{13}\mathrm{C}$ NMR (100 MHz, CDCl₃) δ 209.3, 170.1, 141.0, 121.5, 97.5, 81.1, 77.1, 56.3, 51.0, 50.7, 44.8, 43.2, 41.5, 40.2, 25.3, 22.1, 21.5, 18.4; IR $\nu_{\rm max}$ 2966, 1731, 1372, 1239, 1150, 1035, 710 cm⁻¹; MS (ESI, + ve) m/z 667 [(2M + Na)⁺, 5], 345 [(M + Na)⁺, 100]; HRMS m/z (M + Na)⁺ calcd for C₁₈H₂₆O₅Na 345.1678, found 345.1665.

(3S,3aR,4R,7S,7aS,9R)-9-((tert-Butyldimethylsilyl)oxy)-2,2,4-trimethyl-8-oxo-2,3,3a,4,7,7a-hexahydro-1H-4,7-ethanoinden-3-ylAcetate (47). A magnetically stirred solution of compound 44 (400mg, 1.44 mmol), tert-butyldimethylsilyl chloride (1.73 g, 11.48 mmol),and imidazole (980 mg, 14.39 mmol) in 1,2-dichloroethane (20 mL)kept under a nitrogen atmosphere was heated at reflux for 65 h andthen cooled, quenched with NH₄Cl (10 mL of a saturated aqueoussolution), and extracted with dichloromethane (2 × 40 mL). Thecombined organic phases were washed with water (2 × 40 mL) andbrine (1 × 40 mL) before being dried (Na₂SO₄), filtered, andconcentrated under reduced pressure. The residue thus obtained wassubjected to flash column chromatography (silica, 2:5:100 v/v/v ethyl acetate/dichloromethane/40–60 petroleum ether elution) and gave, after concentration of the appropriate fractions ($R_f = 0.6$ in 3:7 v/v ethyl acetate/40–60 petroleum ether), compound 47 (465 mg, 82%) as a white, crystalline solid: mp = 71–72 °C; [α]_D = +121.0 (*c* 1.0, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ 6.03 (d, J = 8.0 Hz, 1H), 5.87 (t, J = 8.0 Hz, 1H), 5.06 (d, J = 6.0 Hz, 1H), 3.27 (s, 1H), 3.02 (dd, J = 6.0 and 1.8 Hz, 1H), 2.62 (m, 1H), 2.42 (dd, J = 10.6 and 6.0 Hz, 1H), 2.06 (s, 3H), 1.51–1.37 (complex m, 2H), 1.12 (s, 3H), 0.99 (s, 3H), 0.86 (s, 12H), 0.10 (s, 3H), 0.09 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 209.5, 170.3, 141.7, 120.9, 81.4, 76.0, 51.1, 50.5, 44.9, 44.4, 41.7, 40.3, 26.0, 25.5, 22.3, 21.6, 19.0, 18.6, – 4.0, – 5.0; IR ν_{max} 2959, 2930, 2856, 1733, 1237, 1129, 1112, 1093, 858, 837, 778, 709 cm⁻¹; MS m/z 415 [(M + Na)⁺, 100]; HRMS m/z (M + Na)⁺ calcd for C₂₂H₃₆O₄SiNa 415.2281, found 415.2275.

(3S, 3aR, 4R, 7S, 7aS, 9R) - 3 - Acetoxy - 2, 2, 4 - trimethyl - 8 - oxo-2,3,3a,4,7,7a-hexahydro-1H-4,7-ethanoinden-9-yl Benzoate (48). A magnetically stirred solution of compound 44 (2.90 g, 10.42 mmol), 4-(N,N-dimethylamino)pyridine (127 mg, 1.04 mmol), and triethylamine (2.9 mL, 21.00 mmol) in dichloromethane (50 mL) maintained under a nitrogen atmosphere at 0 °C was treated with benzoyl chloride (1.50 mL, 12.50 mmol). The ensuing mixture was allowed to warm to and then stirred at 22 °C for 16 h before being quenched with NH₄Cl (50 mL of a saturated aqueous solution) and then extracted with dichloromethane $(2 \times 100 \text{ mL})$. The combined organic phases were washed with water $(2 \times 100 \text{ mL})$ and then brine $(1 \times 100 \text{ mL})$ before being dried (Na2SO4), filtered, and concentrated under reduced pressure. The residue thus obtained was subjected to flash column chromatography (silica, 2:5:100 v/v/v ethyl acetate/dichloromethane/ 40-60 petroleum ether elution) and gave, after concentration of the appropriate fractions $[R_f = 0.2(5)$ in 1:19 v/v ethyl acetate/40-60 petroleum ether], compound 48 (3.69 g, 93%) as a white, crystalline solid: mp = 134-136 °C; $[\alpha]_{\rm D}$ = +196.0 (c 1.1, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ 7.99 (d, J = 7.3 Hz, 2H), 7.54 (t, J = 7.3 Hz, 1H), 7.41 (t, J = 7.3 Hz, 2H), 6.20 (d, J = 8.1 Hz, 1H), 6.04 (t, J = 8.1 Hz, 1H), 5.12-5.10 (complex m, 2H), 3.23 (dd, J = 6.6 and 1.6 Hz, 1H), 2.80 (m, 1H), 2.72 (dd, J = 10.6 and 6.6 Hz, 1H), 2.07 (s, 3H), 1.55 (dd, J = 12.1 and 7.3 Hz, 1H), 1.43 (m, 1H), 1.11 (s, 3H), 1.02 (s, 3H), 0.87 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 205.0, 170.1, 166.2, 141.3, 133.4, 130.0, 129.5, 128.5, 122.4, 81.1, 74.4, 51.3, 50.7, 44.9, 42.9, 41.7, 39.7, 25.4, 22.2, 21.5, 18.0; IR ν_{max} 2967, 1722, 1268, 1238, 1108, 729, 706 cm⁻¹; MS (ESI, + ve) m/z 787 [(2M + Na)⁺, 100], 405 [(M + Na)⁺, 85]; HRMS m/z (M + Na)⁺ calcd for C23H26O5Na 405.1678, found 405.1682.

(1S,3aS,4S,7R,7aR)-2,2,7-Trimethyl-9-oxo-2,3,3a,4,7,7a-hexahydro-1H-4,7-ethanoinden-1-yl Acetate (49). A magnetically stirred solution of compound 48 (3.30 g, 8.63 mmol) in deoxygenated THF/ methanol (30 mL of a 9:1 v/v mixture) maintained at -78 °C under an argon atmosphere was treated, dropwise over 1 h, with samarium(II) iodide (ca. 250 mL of a 0.086 M solution in THF) at which point a dark-blue color persisted for 0.25 h. The resulting mixture was then quenched with NH₄Cl (100 mL of a saturated aqueous solution) and the ensuing suspension extracted with diethyl ether (2 \times 100 mL). The combined organic phases were washed with HCl $(2 \times 50 \text{ mL of a 1 M aqueous solution})$ and brine $(1 \times 100 \text{ mL})$ before being dried (Na₂SO₄), filtered, and concentrated under reduced pressure. The residue thus obtained was subjected to flash column chromatography (silica, 1:9 v/v diethyl ether/n-hexane elution) and gave, after concentration of the appropriate fractions $[R_f = 0.4(5)]$ in 1:9 v/v diethyl ether/n-hexane)], compound 49 (1.91 g, 85%) as a clear, colorless oil: $[\alpha]_{D} = +161.0$ (c 1.0, CHCl₃); ¹H NMR (400 MHz, $CDCl_3$) δ 6.10 (d, J = 8.1 Hz, 1H), 5.86 (t, J = 8.1 Hz, 1H), 4.93 (d, J = 6.0 Hz, 1H), 2.96 (dd, J = 6.0 and 1.8 Hz, 1H), 2.60 (m, 1H), 2.46 (m, 1H), 1.98 (s, 3H), 1.79 (s, 2H), 1.38 (m, 1H), 1.26 (t, J = 11.5 Hz, 1H), 1.05 (s, 3H), 0.90 (s, 3H), 0.76 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 212.3, 170.0, 142.9, 123.0, 80.4, 53.6, 52.1, 47.7, 44.8, 42.0, 41.2, 39.2, 25.4, 22.2, 22.0, 21.4; IR $\nu_{\rm max}$ 2965, 1726, 1373, 1239, 1069, 707 cm⁻¹; MS (ESI, + ve) m/z 547 [(2M + Na)⁺, 10], 285 [(M + Na)⁺, 100], 263 [(M + H)⁺, 1]; HRMS m/z (M + H)⁺ calcd for C₁₆H₂₃O₃ 263.1647, found 263.1649.

(1R,2aR,4aR,7S,7aR,7bS)-1-(Methoxymethoxy)-6,6,7b-trimethyl-2-oxo-2,2a,4a,5,6,7,7a,7b-octahydro-1H-cyclobuta[e]inden-7-yl Acetate (50). A magnetically stirred solution of compound 45 (100 mg, 0.31 mmol) in dry, deoxygenated dichloromethane (30 mL) maintained under a nitrogen atmosphere was irradiated with a Hanovia 100 W medium-pressure mercury-vapor lamp (maintained at 5 °C) for 16 h. The reaction mixture was then cooled and concentrated under reduced pressure and the residue thus obtained subjected to flash column chromatography (silica, 1:19 v/v ethyl acetate/n-hexane elution) to afford two fractions, A and B.

Concentration of fraction A [$R_f = 0.3(5)$ in 1:19 v/v ethyl acetate/*n*-hexane] gave compound **50** (43 mg, 43% or 99% brsm) as a clear, colorless oil: [α]_D = -157.0 (*c* 0.7, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ 5.76 (m, 1H), 5.73 (m, 1H), 5.54 (m, 1H), 4.97 (d, *J* = 4.2 Hz, 1H), 3.02 (m, 1H), 2.96 (m, 1H), 2.87 (dd, *J* = 9.3 and 4.2 Hz, 1H), 2.14 (s, 3H), 2.06 (m, 1H), 1.98 (s, 3H), 1.51 (dd, *J* = 13.3 and 5.0 Hz, 1H), 1.09 (s, 3H), 1.03 (s, 3H), 0.92 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 201.3, 170.6, 169.9, 134.3, 117.2, 83.4, 83.2, 59.0, 45.4, 43.5, 42.8, 34.8, 34.7, 28.8, 24.3, 21.3, 20.5, 18.4; IR ν_{max} 2933, 2873, 1792, 1735, 1373, 1224, 1087 cm⁻¹; MS (ESI, + ve) *m/z* 375 [(M + MeOH + Na)⁺, 20], 343 [(M + Na)⁺, 100]; HRMS *m/z* (M + Na)⁺ calcd for C₁₈H₂₄O₅Na 343.1521, found 343.1520.

Concentration of the fraction B [$R_f = 0.2(5)$ in 1:19 v/v ethyl acetate/*n*-hexane] gave compound **45** (56 mg, 56% recovery) as a colorless, crystalline solid that was identical, in all respects, with an authentic sample.

(1R,2aR,4aR,7S,7aR,7bS)-1-(Methoxymethoxy)-6,6,7b-trimethyl-2-oxo-2,2a,4a,5,6,7,7a,7b-octahydro-1H-cyclobuta[e]inden-7-yl Acetate (51). A magnetically stirred solution of compound 46 (100 mg, 0.31 mmol) in dry, deoxygenated dichloromethane (30 mL) maintained under a nitrogen atmosphere was irradiated with a Hanovia 100 W medium-pressure mercury-vapor lamp (maintained at 5 °C) for 16 h. The cooled reaction mixture was concentrated under reduced pressure and the residue thus obtained subjected to flash column chromatography (silica, 1:9 v/v ethyl acetate/n-hexane elution), thus affording two fractions, A and B.

Concentration of the fraction A [$R_f = 0.3(5)$ in 1:9 v/v ethyl acetate/*n*-hexane] gave compound **51** (27 mg, 27% or 98% brsm) as a clear, colorless oil: [α]_D = -324.0 (*c* 0.2, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ 5.70 (m, 1H), 5.55 (m, 1H), 5.00 (d, *J* = 4.1 Hz, 1H), 4.67 (m, 3H), 3.42 (s, 3H), 2.90 (m, 1H), 2.85 (m, 1H), 2.76 (dd, *J* = 9.4 and 4.1 Hz, 1H), 2.05 (m, 1H), 1.98 (s, 3H), 1.51 (dd, *J* = 13.2 and 5.2 Hz, 1H), 1.11 (s, 3H), 1.06 (s, 3H), 0.93 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 204.5, 170.7, 133.4, 118.0, 97.3, 89.4, 83.2, 58.5, 56.1, 45.4, 43.5, 43.2, 34.7, 34.4, 28.7, 24.2, 21.4, 18.5; IR ν_{max} 2962, 2933, 1787, 1733, 1373, 1240, 1110, 1067, 1012 cm⁻¹; MS (ESI, + ve) *m*/*z* 345 [(M + Na)⁺, 100]; HRMS *m*/*z* (M + Na)⁺ calcd for C₁₈H₂₆O₅Na 345.1678, found 345.1672.

Concentration of the fraction B [$R_f = 0.2(5)$ in 1:9 v/v ethyl acetate/*n*-hexane] gave compound **46** (71 mg, 71% recovery) as a clear, colorless oil that was identical, in all respects, with an authentic sample.

(3S,3aR,4R,7S,7aS,9R)-9-((tert-Butyldimethylsilyl)oxy)-2,2,4-trimethyl-8-oxo-2,3,3a,4,7,7a-hexahydro-1H-4,7-ethanoinden-3-yl Acetate (52). A magnetically stirred solution of compound 47 (100 mg, 0.25 mmol) in dry, deoxygenated dichloromethane (30 mL) maintained under a nitrogen atmosphere was irradiated with a Hanovia 100 W medium-pressure mercury-vapor lamp (maintained at 5 °C) for 16 h. The cooled reaction mixture was concentrated under reduced pressure and the residue thus obtained subjected to flash column chromatography (silica, 1:19 v/v ethyl acetate/n-hexane elution), thus thereby affording two fractions, A and B.

Concentration of the fraction A ($R_f = 0.4$ in 1:19 v/v ethyl acetate/ *n*-hexane) gave compound **52** (18 mg, 18% or 99% brsm) as a clear, colorless oil: [α]_D = -189.0 (*c* 2.3, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ 5.64 (m, 1H), 5.55 (m, 1H), 4.98 (d, *J* = 4.1 Hz, 1H), 4.71 (d, *J* = 2.8 Hz, 1H), 2.86–2.78 (complex m, 2H), 2.65 (dd, *J* = 9.3 and 4.1 Hz, 1H), 2.03 (m, 1H), 1.97 (s, 3H), 1.50 (dd, *J* = 13.2 and 5.1 Hz, 1H), 1.10 (s, 3H), 0.97 (s, 3H), 0.92 (s, 3H), 0.90 (s, 9H), 0.10 (s, 3H), 0.07 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 206.1, 170.6, 132.8, 118.7, 85.0, 83.3, 58.0, 45.5, 43.5, 43.2, 34.8, 34.5, 28.8, 25.8, 24.2, 21.3, 18.4, 18.2, - 4.6, - 4.9; IR $\nu_{\rm max}$ 2958, 2931, 2859, 1788, 1736, 1239, 1108, 883, 838, 780 cm $^{-1}$; MS (ESI, + ve) m/z 415 [(M + Na)+, 100], 393 [(M + H)+, 5]; HRMS m/z (M + Na)+ calcd for $\rm C_{22}H_{36}O_4SiNa$ 415.2281, found 415.2284.

Concentration of the fraction B ($R_f = 0.3$ in 1:19 v/v ethyl acetate/ *n*-hexane) gave compound 47 (81 mg, 81% recovery) as a colorless, crystalline solid that was identical, in all respects, with an authentic sample.

(1R,2aR,4aR,7S,7aR,7bS)-7-Acetoxy-6,6,7b-trimethyl-2-oxo-2,2a,4a,5,6,7,7a,7b-octahydro-1H-cyclobuta[e]inden-1-yl Benzoate (53). A magnetically stirred solution of compound 48 (100 mg, 0.26 mmol) in dry, deoxygenated dichloromethane (30 mL) maintained under a nitrogen atmosphere was irradiated with a Hanovia 100 W medium-pressure mercury-vapor lamp (maintained at 5 °C) for 16 h. The reaction mixture was then cooled and concentrated under reduced pressure and the residue so obtained subjected to flash column chromatography (silica, 1:19 v/v ethyl acetate/n-hexane elution), thus affording two fractions, A and B.

Concentration of the fraction A [$R_f = 0.3(5)$ in 1:19 v/v ethyl acetate/*n*-hexane] gave compound **53** (17 mg, 17% or 100% brsm) as a colorless, crystalline solid: mp = 180–181 °C; [α]_D = -672.0 (*c* 1.5, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ 8.06 (m, 2H), 7.59 (t, *J* = 7.4 Hz, 1H), 7.45 (t, *J* = 7.4 Hz, 2H), 5.97 (d, *J* = 2.8 Hz, 1H), 5.80 (m, 1H), 5.59 (m, 1H), 4.98 (d, *J* = 4.2 Hz, 1H), 3.10 (m, 1H), 3.04 (m, 1H), 2.97 (dd, *J* = 9.3 and 4.2 Hz, 1H), 2.08 (dd, *J* = 13.3 and 9.8 Hz, 1H), 1.99 (s, 3H), 1.53 (dd, *J* = 13.3 and 5.0 Hz, 1H), 1.12 (s, 3H), 1.10 (s, 3H), 0.92 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 201.1, 170.6, 165.3, 134.3, 133.8, 130.1, 128.8, 128.7, 117.3, 83.6, 83.2, 59.2, 45.4, 43.5, 42.9, 35.1, 34.8, 28.8, 24.3, 21.3, 18.5; IR ν_{max} 2963, 2934, 2872, 1790, 1724, 1372, 1237, 1104, 709 cm⁻¹; MS (ESI, + ve) *m*/*z* 787 [(2M + Na)⁺, 10], 437 [(M + MeOH + Na)⁺, 50], 405 [(M + Na)⁺, 100]; HRMS *m*/*z* (M + Na)⁺ calcd for C₂₃H₂₆O₅Na 405.1678, found 405.1670.

Concentration of the fraction B [$R_f = 0.2(5)$ in 1:19 v/v ethyl acetate/*n*-hexane] gave compound **48** (83 mg, 83% recovery) as a colorless, crystalline solid that was identical, in all respects, with an authentic sample.

(2 a R, 4 a R, 7 S, 7 a R, 7 b S) - 6, 6, 7 b - Trimethyl-2-oxo-2,2a,4a,5,6,7,7a,7b-octahydro-1H-cyclobuta[e]inden-7-yl Acetate (54). A magnetically stirred solution of compound 49 (100 mg, 0.38 mmol) in dry, deoxygenated dichloromethane (30 mL) maintained under a nitrogen atmosphere was irradiated with a Hanovia 100 W medium-pressure mercury-vapor lamp (maintained at 5 °C) for 16 h. The reaction mixture was then cooled and concentrated under reduced pressure and the ensuing residue subjected to flash column chromatography (silica, 1:19 v/v ethyl acetate/n-hexane elution), thus affording two fractions, A and B.

Concentration of the fraction A [$R_f = 0.3(5)$ in 1:19 v/v ethyl acetate/*n*-hexane] gave compound **54** (23 mg, 23% or 92% brsm) as a clear, colorless oil, [α]_D = -254.0 (*c* 3.4, CHCl₃). ¹H NMR (400 MHz, CDCl₃) δ 5.73 (m, 1H), 5.48 (m, 1H), 4.97 (d, *J* = 4.2 Hz, 1H), 3.12 (dd, *J* = 16.2 and 2.1 Hz, 1H), 3.05 (m, 1H), 2.84 (m, 1H), 2.59 (dd, *J* = 9.2 and 4.3 Hz, 1H), 2.51 (dd, *J* = 16.2 and 6.0 Hz, 1H), 2.01 (dd, *J* = 13.2 and 9.7 Hz, 1H), 1.95 (s, 3H), 1.50 (dd, *J* = 13.2 and 4.8 Hz, 1H), 1.19 (s, 3H), 1.08 (s, 3H), 0.90 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 206.4, 170.7, 133.7, 118.2, 83.5, 64.1, 56.7, 45.7, 44.3, 43.4, 34.9, 29.0, 28.1, 25.5, 24.3, 21.3; IR ν_{max} 3020, 2966, 2871, 1778, 1731, 1373, 1239, 1020, 711 cm⁻¹; MS (ESI, + ve) *m*/*z* 285 [(M + Na)⁺, 100]; HRMS *m*/*z* (M + Na)⁺ calcd for C₁₆H₂₂O₃Na 285.1467, found 285.1472.

Concentration of the fraction B ($R_f = 0.3$ in 1:19 v/v ethyl acetate/ *n*-hexane) gave compound **49** (69 mg, 69% recovery) as a colorless, crystalline solid that was identical, in all respects, with an authentic sample.

(1*S*,1*aR*,3*aR*,6*S*,6*aR*,6*bR*)-5,5,6*b*-Trimethyl-1,1*a*,3*a*,4,5,6,6*a*,6*b*octahydrocyclopropa[e]indene-1,6-diyl Diacetate (55). A magnetically stirred solution of compound 50 (40 mg, 0.12 mmol) in dry, deoxygenated dichloromethane (10 mL) maintained under a nitrogen atmosphere was irradiated with a Hanovia 450 W medium-pressure mercury-vapor lamp for 0.5 h. The reaction mixture was then cooled and concentrated under reduced pressure and the residue thus obtained subjected to flash column chromatography (silica, 1:9 v/v ethyl acetate/n-hexane elution). Concentration of the appropriate fractions $[R_t = 0.6(5)$ in 1:9 v/v ethyl acetate/n-hexane] gave compound 55 (39 mg, 98%) as a clear, colorless oil: $[\alpha]_D = +167.0$ (c 0.5, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ 5.90 (m, 1H), 5.22 (dd, J = 10.0 and 1.9 Hz, 1H), 5.04 (d, J = 3.9 Hz, 1H), 3.77 (d, J = 2.3 Hz, 1H), 2.82 (dd, J = 10.1 and 3.9 Hz, 1H), 2.64 (m, 1H), 2.05 (s, 3H), 2.00 (s, 3H), 1.90 (dd, J = 13.1 and 9.6 Hz, 1H), 1.33 (dd, J = 13.1 and 5.2 Hz, 1H), 1.22 (dd, J = 11.1 and 2.3 Hz, 1H), 1.06 (s, 6H), 0.90 (s, 3H); 13 C NMR (100 MHz, CDCl₃) δ 171.8, 171.1, 129.9, 123.1, 84.0, 63.4, 46.6, 42.8, 40.8, 35.4, 28.6, 24.3, 24.2, 22.1, 21.3, 20.9, 15.9; IR $\nu_{\rm max}$ 3026, 2966, 2933, 2872, 1733, 1371, 1227 cm⁻¹; MS (ESI, + ve) m/z 347 [(M + MeOH + Na)⁺, 100%], 331 [(M + K)⁺, 20], 329 (50), 315 [(M + Na)⁺, 60]; HRMS m/z (M + Na)⁺ calcd for C₁₇H₂₄O₄Na 315.1572, found 315.1570.

(15,1aR,3aR,6S,6aR,6bS)-1-((tert-Butyldimethylsilyl)oxy)-5,5,6btrimethyl-1,1a,-3a,4,5,6,6a,6b-octahydrocyclopropa[e]inden-6-yl Acetate (56). A magnetically stirred solution of compound 47 (100 mg, 0.25 mmol) in dry, deoxygenated dichloromethane (30 mL) maintained under a nitrogen atmosphere was irradiated with a Hanovia 450 W medium-pressure mercury-vapor lamp for 1.67 h. The reaction mixture was then cooled, concentrated under reduced pressure and the residue thus obtained subjected to flash column chromatography (silica, 1:19 v/v ethyl acetate/40–60 petroleum ether) to afford two fractions, A and B.

Concentration of the fraction A ($R_f = 0.6$ in 1:4 v/v ethyl acetate/ 40–60 petroleum ether) gave compound **56** (74 mg, 80% or 87% brsm) as a clear, colorless oil: [α]_D = -65.5 (*c* 1.0, CHCl₃); ¹H NMR [400 MHz, (CD₃)₂CO] δ 5.92 (m, 1H), 5.11 (dd, *J* = 10.0 and 2.1 Hz, 1H), 5.01 (d, *J* = 4.1 Hz, 1H), 3.14 (d, *J* = 2.2 Hz, 1H), 2.78 (m, 1H), 2.63 (m, 1H), 1.95 (s, 3H), 1.89 (m, 1H), 1.30 (dd, *J* = 13.0 and 5.2 Hz, 1H), 1.09 (s, 3H), 1.06 (s, 3H), 0.97 (dd, *J* = 5.5 and 2.4 Hz, 1H), 0.90 (s, 9H), 0.87 (s, 3H), 0.10(1) (s, 3H), 0.09(8) (s, 3H); ¹³C NMR [100 MHz, (CD₃)₂CO] δ 170.7, 128.9, 125.2, 84.4, 63.9, 47.3, 43.3, 41.9, 36.2, 28.9, 27.7, 26.2, 24.7, 23.1, 21.1, 18.6, 15.9, - 4.9 (one signal obscured or overlapping); IR ν_{max} 3021, 2958, 2930, 2859, 1734, 1240, 1153, 859, 835, 776 cm⁻¹; MS (ESI, + ve) *m*/*z* 387 [(M + Na)⁺, 100]; HRMS *m*/*z* (M + Na)⁺ calcd for C₂₁H₃₆O₃SiNa 387.2331, found 387.2331.

Concentration of the fraction B ($R_f = 0.5$ in 1:4 v/v ethyl acetate/ 40–60 petroleum ether) gave compound 47 (8 mg, 8% recovery) as a white, crystalline solid that was identical, in all respects, with an authentic sample.

(3S, 3aR, 4R, 7S, 7aS, 9R)-2, 2, 4-Trimethyl-9-((methylsulfonyl)oxy)-8oxo-2,3,3a,4,-7,7a-hexahydro-1H-4,7-ethanoinden-3-yl Acetate (57). A magnetically stirred solution of compound 44 (280 mg, 1.01 mmol), methanesulfonyl chloride (94 µL, 1.21 mmol) and triethylamine (212 μ L, 1.52 mmol) in dichloromethane (5 mL) maintained under a nitrogen atmosphere was stirred at 22 °C for 48 h and then quenched with NH₄Cl (10 mL of a saturated aqueous solution) before being extracted with dichloromethane $(2 \times 50 \text{ mL})$. The combined organic fractions were washed with water $(2 \times 50 \text{ mL})$ and brine $(1 \times 10^{-1} \text{ mL})$ 50 mL) and then dried (MgSO₄), filtered, and concentrated under reduced pressure. The residue thus obtained was subjected to flash column chromatography (silica, 1:9 v/v ethyl acetate/40-60 petroleum ether elution) to give, after concentration of the appropriate fractions ($R_f = 0.3$ in 1:9 v/v ethyl acetate/40–60 petroleum ether), compound 57 (198 mg, 56%) as a light-yellow oil: $[\alpha]_D = +114.0$ (c 4.3, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ 6.08 (d, J = 8.1 Hz, 1H), 5.95 (t, J = 8.1 Hz, 1H), 5.08 (d, J = 5.9 Hz, 1H), 4.31 (s, 1H), 3.19 (m, 1H), 3.17 (s, 3H), 2.69 (m, 1H), 2.58 (dd, J = 10.6 and 6.0 Hz, 1H), 2.05 (s, 3H), 1.53 (m, 1H), 1.39 (t, J = 12.0 Hz, 1H), 1.20 (s, 3H), 0.98 (s, 3H), 0.85 (s, 3H); $^{13}\mathrm{C}$ NMR (100 MHz, CDCl₃) δ 204.1, 170.0, 140.9, 121.9, 81.3, 81.0, 50.7, 50.2, 44.9, 42.7, 41.6, 39.5(1), 39.4(8), 25.3, 22.1, 21.5, 18.1; IR $\nu_{\rm max}$ 2968, 1730, 1358, 1238, 1173, 965, 841 cm⁻¹; MS (ESI, + ve) m/z 735 [(2M + Na)⁺, 15], 379 [(M + Na)⁺, 100]; HRMS m/z (M + Na)⁺ calcd for C17H24O6SNa 379.1191, found 379.1193.

 $(1R,2a^{1}S,2cR,5S,5aR,5bR)-4,4,5b$ -Trimethyl-1-((methylsulfonyl)oxy)-2-oxodeca-hydro-1H-cyclopenta[a]cyclopropa[cd]pentalen-5yl Acetate (58). A magnetically stirred solution of compound 57 (100 mg, 0.28 mmol) in dry, deoxygenated dichloromethane (30 mL) maintained under a nitrogen atmosphere was irradiated with a Hanovia 100 W medium-pressure mercury-vapor lamp (maintained at 5 °C) for 12 h. The reaction mixture was then cooled and concentrated under reduced pressure and the residue thus obtained subjected to flash column chromatography (silica, 1:9 v/v ethyl acetate/n-hexane elution) to afford two fractions, A and B.

Concentration of the fraction A [$R_f = 0.4(5)$ in 1:9 v/v ethyl acetate/*n*-hexane] gave compound **58** (48 mg, 48% or 98% brsm) as a clear, colorless oil: [α]_D = +161.0 (*c* 1.0, CHCl₃). ¹H NMR (400 MHz, CDCl₃) δ 5.54 (*s*, 1H), 5.09 (d, *J* = 4.8 Hz, 1H), 3.05 (*s*, 3H), 2.85–2.72 (complex m, 2H), 2.50 (d, *J* = 4.8 Hz, 1H), 2.23 (*s*, 3H), 2.18 (*s*, 1H), 1.71 (d, *J* = 4.8 Hz, 1H), 1.59 (dd, *J* = 8.7 and 2.3 Hz, 2H), 1.07 (*s*, 3H), 0.97(2) (*s*, 3H), 0.96(7) (*s*, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 211.1, 170.2, 80.9, 78.8, 50.4, 45.2, 42.5, 42.4, 41.1, 39.3, 37.6, 36.1, 32.9, 25.1, 22.7, 22.5, 21.2; IR ν_{max} 2966, 2936, 2875, 1729, 1356, 1236, 1172, 917, 842 cm⁻¹; MS *m*/*z* 735 [(2M + Na)⁺, 5], 379 [(M + Na)⁺, 100]; HRMS *m*/*z* (M + Na)⁺ calcd for C₁₇H₂₄O₆SNa 379.1191, found 379.1189.

Concentration of the fraction B ($R_f = 0.3$ in 1:9 v/v ethyl acetate/*n*-hexane) gave compound **57** (50 mg, 50% recovery) as a clear, colorless oil that was identical, in all respects, with an authentic sample.

X-ray Crystallographic Studies. *Crystallographic Data. Compound* **8**: $C_9H_{12}O_3$, M = 168.19, T = 150 K, orthorhombic, space group $P2_12_12_1$, Z = 4, a = 7.3238(1) Å, b = 7.5553(1) Å, c = 14.5389(2) Å; V = 804.49(2) Å³, $D_x = 1.389$ g cm⁻³, 1585 unique data $(2\theta_{max} = 144.6^\circ)$, R = 0.023 [for 1569 reflections with $I > 2.0\sigma(I)$]; $R_w = 0.057$ (all data), S = 1.00.

Compound 9: $C_{15}H_{26}O_3$ Si, M = 282.46, T = 150 K, monoclinic, space group C2, Z = 4, a = 18.9377(5) Å, b = 6.2442(1) Å, c = 14.6275(5) Å; $\beta = 105.950(3)^\circ$; V = 1663.12(8) Å³, $D_x = 1.128$ g cm⁻³, 3042 unique data ($2\theta_{max} = 144.8^\circ$), R = 0.032 [for 2889 reflections with $I > 2.0\sigma(I)$]; $R_w = 0.080$ (all data), S = 0.99.

Compound 10: $C_9H_{12}O_3$, M = 168.19, T = 150 K, orthorhombic, space group $P_{2_12_12_1}$, Z = 8, a = 8.23350(7) Å, b = 8.68949(8) Å, c = 23.6396(2) Å; V = 1691.29(3) Å³, $D_x = 1.321$ g cm⁻³, 3334 unique data $(2\theta_{max} = 144.6^{\circ})$, R = 0.04 [for 3257 reflections with $I > 2.0\sigma(I)$]; $R_w = 0.112$ (all data), S = 1.01.

Compound **13**: $C_{14}H_{26}O_2S_i$, M = 254.45, T = 150 K, orthorhombic, space group $P2_12_12$, Z = 8, a = 18.3637(1) Å, b = 14.1093(1) Å, c = 12.0122(1) Å; V = 3112.35(4) Å³, $D_x = 1.086$ g cm⁻³, 6159 unique data ($2\theta_{max} = 145.0^{\circ}$), R = 0.025 [for 6101 reflections with $I > 2.0\sigma(I)$]; $R_w = 0.067$ (all data), S = 1.01.

Compound 23: $C_{13}H_{16}O_3$, M = 220.27, T = 150 K, orthorhombic, space group $P2_12_12_1$, Z = 4, a = 7.27015(5) Å, b = 11.05109(8) Å, c = 13.78627(9) Å; V = 1107.63(1) Å³, $D_x = 1.321$ g cm⁻³, 2186 unique data $(2\theta_{max} = 144.8^{\circ})$, R = 0.022 [for 2152 reflections with $I > 2.0\sigma(I)$]; $R_w = 0.057$ (all data), S = 1.01.

Compound **27**: $C_{13}H_{16}O_3$, M = 220.27, T = 200 K, monoclinic, space group $P2_1$, Z = 2, a = 7.2003(2) Å, b = 8.4682(2) Å, c = 9.4258(3) Å; $\beta = 91.1279(17)^\circ$; V = 574.61(3) Å³, $D_x = 1.273$ g cm⁻³, 1778 unique data ($2\theta_{max} = 60.0^\circ$), R = 0.048 [for 1655 reflections with $I > 2.0\sigma(I)$]; $R_w = 0.156$ (all data), S = 1.03.

Compound **30**: $C_{11}H_{12}O_3$, M = 192.21, T = 150 K, monoclinic, space group $P2_1$, Z = 2, a = 8.4949(4) Å, b = 5.9702(2) Å, c = 9.3840(4) Å; $\beta = 95.383(4)^\circ$; V = 473.82(3) Å³, $D_x = 1.347$ g cm⁻³, 1032 unique data ($2\theta_{max} = 144.8^\circ$), R = 0.045 [for 998 reflections with $I > 2.0\sigma(I)$]; $R_w = 0.130$ (all data), S = 1.03.

Compound **32**: $C_{10}H_{12}O_2$, M = 164.20, T = 150 K, orthorhombic, space group $P2_12_12_1$, Z = 4, a = 7.8054(1) Å, b = 9.7920(1) Å, c = 11.0580(1) Å; V = 845.17(2) Å³, $D_x = 1.290$ g cm⁻³, 1677 unique data $(2\theta_{max} = 144.6^\circ)$, R = 0.022 [for 1645 reflections with $I > 2.0\sigma(I)$]; $R_w = 0.057$ (all data), S = 1.01.

Compound **34**: $C_{11}H_{12}O_3$, M = 192.21, T = 150 K, monoclinic, space group $P2_1$, Z = 2, a = 6.9303(2) Å, b = 7.7696(2) Å, c = 8.6272(2) Å; $\beta = 92.403(2)^\circ$; V = 464.13(2) Å³, $D_x = 1.375$ g cm⁻³,

993 unique data $(2\theta_{\text{max}} = 145.2^{\circ})$, R = 0.037 [for 980 reflections with $I > 2.0\sigma(I)$]; $R_{w} = 0.117$ (all data), S = 1.04.

Compound **35**: $C_{13}H_{16}O_3$, M = 220.27, T = 150 K, orthorhombic, space group $P2_12_12_1$, Z = 4, a = 7.3877(1) Å, b = 7.8178(1) Å, c = 19.8494(2) Å; V = 1146.41(2) Å³, $D_x = 1.276$ g cm⁻³, 2267 unique data ($2\theta_{max} = 144.8^{\circ}$), R = 0.030 [for 2238 reflections with $I > 2.0\sigma(I)$]; $R_w = 0.056$ (all data), S = 1.00.

Compound **36**: $C_{11}H_{12}O_3$, M = 192.21, T = 150 K, orthorhombic, space group $P2_12_12_1$, Z = 4, a = 7.4666(2) Å, b = 12.0831(5) Å, c = 10.8645(4) Å; V = 980.19(6) Å³, $D_x = 1.302$ g cm⁻³, 1491 unique data ($2\theta_{max} = 58.4^{\circ}$), R = 0.047 [for 1303 reflections with $I > 2.0\sigma(I)$]; $R_w = 0.108$ (all data), S = 1.00.

Compound **38**: $C_{10}H_{12}O_2$, M = 164.20, T = 150 K, hexagonal, space group $P6_5$, Z = 6, a = 14.7787(4) Å, c = 7.6213(2) Å; V = 1441.56(7) Å³, $D_x = 1.135$ g cm⁻³, 1379 unique data $(2\theta_{max} = 58.8^\circ)$, R = 0.045 [for 1283 reflections with $I > 2.0\sigma(I)$]; $R_w = 0.111$ (all data), S = 1.03.

Compound 41: $C_{17}H_{26}O_3$, M = 278.39, T = 150 K, orthorhombic, space group $P2_12_12_1$, Z = 4, a = 6.26451(4) Å, b = 12.09608(9) Å, c = 20.26568(14) Å; V = 1535.65(2) Å³, $D_x = 1.204$ g cm⁻³, 3025 unique data ($2\theta_{max} = 144.6^{\circ}$), R = 0.024 [for 2974 reflections with $I > 2.0\sigma(I)$]; $R_w = 0.062$ (all data), S = 1.00.

Compound 44: $C_{16}H_{22}O_4$, M = 278.35, T = 150 K, orthorhombic, space group $P2_12_12_1$, Z = 4, a = 6.4307(1) Å, b = 8.7685(1) Å, c = 26.1929(2) Å; V = 1476.95(3) Å³, $D_x = 1.252$ g cm⁻³, 2913 unique data $(2\theta_{max} = 144.8^{\circ})$, R = 0.024 [for 2859 reflections with $I > 2.0\sigma(I)$]; $R_w = 0.063$ (all data), S = 1.00.

Compound **45**: $C_{18}H_{24}O_5$, M = 320.39, T = 150 K, monoclinic, space group $P2_1$, Z = 2, a = 7.4300(3) Å, b = 9.7667(3) Å, c = 12.1277(3) Å; $\beta = 92.279(3)^\circ$; V = 879.37(5) Å³, $D_x = 1.210$ g cm⁻³, 2377 unique data ($2\theta_{max} = 58.6^\circ$), R = 0.038 [for 2242 reflections with $I > 2.0\sigma(I)$]; $R_w = 0.093$ (all data), S = 1.02.

Compound 47: $C_{22}H_{36}O_4$ Si, M = 392.61, T = 150 K, orthorhombic, space group $P2_12_12_1$, Z = 4, a = 25.6494(7) Å, b = 14.1949(4) Å, c = 6.2343(2) Å; V = 2269.85(11) Å³, $D_x = 1.149$ g cm⁻³, 5914 unique data $(2\theta_{max} = 58.8^{\circ})$, R = 0.045 [for 5443 reflections with $I > 2.0\sigma(I)$]; $R_w = 0.102$ (all data), S = 1.04.

Compound **48**: $C_{23}H_{26}O_5$, M = 382.46, T = 150 K, orthorhombic, space group $P2_12_12_1$, Z = 4, a = 19.9195(6) Å, b = 11.6626(3) Å, c = 8.7386(2) Å; V = 2030.09(9) Å³, $D_x = 1.251$ g cm⁻³, 2980 unique data ($2\theta_{max} = 58.6^{\circ}$), R = 0.041 [for 2759 reflections with $I > 2.0\sigma(I)$]; $R_w = 0.101$ (all data), S = 1.03.

Compound **53**: $C_{23}H_{26}O_5$, M = 382.46, T = 150 K, orthorhombic, space group $P2_12_12_1$, Z = 4, a = 8.9344(2) Å, b = 15.4099(3) Å, c = 14.9268(3) Å; V = 2055.09(7) Å³, $D_x = 1.236$ g cm⁻³, 3065 unique data $(2\theta_{max} = 59.0^{\circ})$, R = 0.042 [for 2868 reflections with $I > 2.0\sigma(I)$]; $R_w = 0.112$ (all data), S = 1.04.

Structure Determination. Images for compound 27, 36, 38, 45, 47, 48, and 53 were measured on a diffractometer (Mo K α , graphite monochromator, $\lambda = 0.71073$ Å) fitted with an area detector and the data extracted using the DENZO/Scalepack or CrysAlis package.¹⁵ Images for compounds 8, 9, 10, 13, 23, 30, 32, 34, 35, 41, and 44 were measured on a diffractometer (Cu K α , mirror monochromator, λ = 1.54184 Å) fitted with an area detector and the data extracted using the CrysAlis package.¹⁶ The structure solutions for all six compounds were solved by direct methods (SIR92)¹⁷ then refined using the CRYSTALS program package.¹⁸ Atomic coordinates, bond lengths and angles, and displacement parameters have been deposited at the Cambridge Crystallographic Data Centre (CCDC nos. 1545223, 154224, 1545225, 1545226, 1545227, 1545228, 1545229, 1545230, 1545231, 1545232, 1545233, 1545234, 1545235, 1545236, 1545237, 1545238, 1545239 and 1545240). These data can be obtained free of charge via www.ccdc.cam.ac.uk/data_request/cif, by emailing data_ request@ccdc.cam.ac.uk, or by contacting the Cambridge Crystallographic Data Centre, 12 Union Road, Cambridge CB2 1EZ, UK; fax: + 44 1223 336033.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.joc.7b01243.

Anisotropic displacement ellipsoid plots from the singlecrystal X-ray analyses of compounds 8–10, 13, 23, 27, 30, 32, 34–36, 38, 41, 44, 45, 47, 48, and 53; ¹H and ¹³C NMR spectra of compounds 8–39 and 42– 58(PDF)

X-ray crystallographic data for compounds 8–10, 13, 23, 27, 30, 32, and 34 (CIF)

X-ray crystallographic data for compounds 35, 36, 38, 41, 44, 45, 47, 48, and 53 (CIF)

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Notes

The authors declare no competing financial interest.

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REFERENCES

(1) For a relevant and recent review, see: Banwell, M. G.; Bon, D. J.-Y. D. Applications of the Di- π -Methane and Related Rearrangement Reactions in Chemical Synthesis. In *Molecular Rearrangements in Organic Synthesis*; Rojas, C. M., Ed.; Wiley: Hoboken, NJ, 2015; Chapter 9, pp 261–288.

(2) Luo, S.-Y.; Jang, Y.-J.; Liu, J.-Y.; Chu, C.-S.; Liao, C.-C.; Hung, S.-C. Angew. Chem., Int. Ed. 2008, 47, 8082 and references cited therein.
(3) For reviews on cis-1,2-dihydrocatechols, see: (a) Johnson, R. A. Org. React. 2004, 63, 117. (b) Hudlicky, T.; Reed, J. W. Synlett 2009, 2009, 685. (c) Lewis, S. E. Chem. Commun. 2014, 50, 2821. (d) Banwell, M. G.; Bolte, B.; Buckler, J. N.; Chang, E. L.; Lan, P.; Taher, E. S.; White, L. V.; Willis, A. C. J. Proc. R. Soc. New South Wales 2016, 149, 34.

(4) (a) Banwell, M. G.; Edwards, A. J.; Harfoot, G. J.; Jolliffe, K. A. *Tetrahedron* 2004, 60, 535. (b) Banwell, M. G.; Austin, K. A. B.; Willis, A. C. *Tetrahedron* 2007, 63, 6388. (c) Reekie, T. A.; Austin, K. A. B.; Banwell, M. G.; Willis, A. C. Aust. J. Chem. 2008, 61, 94. (d) Bon, D. J.-Y. D.; Banwell, M. G.; Willis, A. C. *Tetrahedron* 2010, 66, 7807. (e) Bon, D. J.-Y. D.; Banwell, M. G.; Cade, I. A.; Willis, A. C. *Tetrahedron* 2011, 67, 8348. (f) Bon, D. J.-Y. D.; Banwell, M. G.; Willis, A. C. Tetrahedron 2013, 69, 1363.

(5) (a) Schwartz, B. D.; Matoušová, E.; White, R.; Banwell, M. G.; Willis, A. C. *Org. Lett.* **2013**, *15*, 1934. (b) Chang, E. L.; Bolte, B.; Lan, P.; Willis, A. C.; Banwell, M. G. J. Org. Chem. **2016**, *81*, 2078.

(6) Lan, P.; Banwell, M. G.; Willis, A. C. Org. Lett. 2015, 17, 166.

(7) Yan, Q.; Banwell, M. G.; Coote, M. L.; Lee, R.; Willis, A. C. Chem. - Asian J. 2017, 12, 1480.

(8) Banwell, M. G.; Chun, C.; Edwards, A. J.; Vögtle, M. M. Aust. J. Chem. 2003, 56, 861.

(9) Ireland, R. E.; Liu, L. J. Org. Chem. 1993, 58, 2899.

(10) This protocol is based on one first described by Ma and Bobbitt: Ma, Z.; Bobbitt, J. M. *J. Org. Chem.* **1991**, *56*, 6110. For examples of related oxidations see refs 4b-d.

(11) Ravelli, D.; Fagnoni, M. Photochemistry of Phosphate and Sulfonate Esters In CRC Handbook of Organic Photochemistry, 3rd ed.; Griesbeck, A., Oelgemoller, M., Ghetti, F., Ed.; CRC Press, Boca Raton, FL, 2012; Chapter 17, pp 393–417.

(12) Still, W. C.; Kahn, M.; Mitra, A. J. Org. Chem. 1978, 43, 2923. (13) Pangborn, A. B.; Giardello, M. A.; Grubbs, R. H.; Rosen, R. K.; Timmers, F. J. Organometallics 1996, 15, 1518.

(14) Foley, D. A.; O'Leary, P.; Buckley, N. R.; Lawrence, S. E.; Maguire, A. R. *Tetrahedron* **2013**, *69*, 1778.

(15) DENZO-SMN: Otwinowski, Z.; Minor, W. Processing of X-ray diffraction data collected in oscillation mode. In *Methods in Enzymology*; Carter, C. W., Jr., Sweet, R. M., Eds.; Academic Press: New York, 1997; Vol. 276: Macromolecular Crystallography, Part A, pp 307–326.

(16) CrysAlis PRO, version 1.171.37.35h (release 09-02-2015 CrysAlis171.NET) (compiled Feb 9 2015, 16:26:32) Agilent Technologies: Oxfordshire, UK, 2015.

(17) SIR92: Altomare, A.; Cascarano, G.; Giacovazzo, C.; Guagliardi, A.; Burla, M. C.; Polidori, G.; Camalli, M. *J. Appl. Crystallogr.* **1994**, *27*, 435.

(18) Betteridge, P. W.; Carruthers, J. R.; Cooper, R. I.; Prout, K.; Watkin, D. J. J. Appl. Crystallogr. 2003, 36, 1487.